
H₂ in Space[©]

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October 2014



H₂ in Space

[H₂ Structure](#)

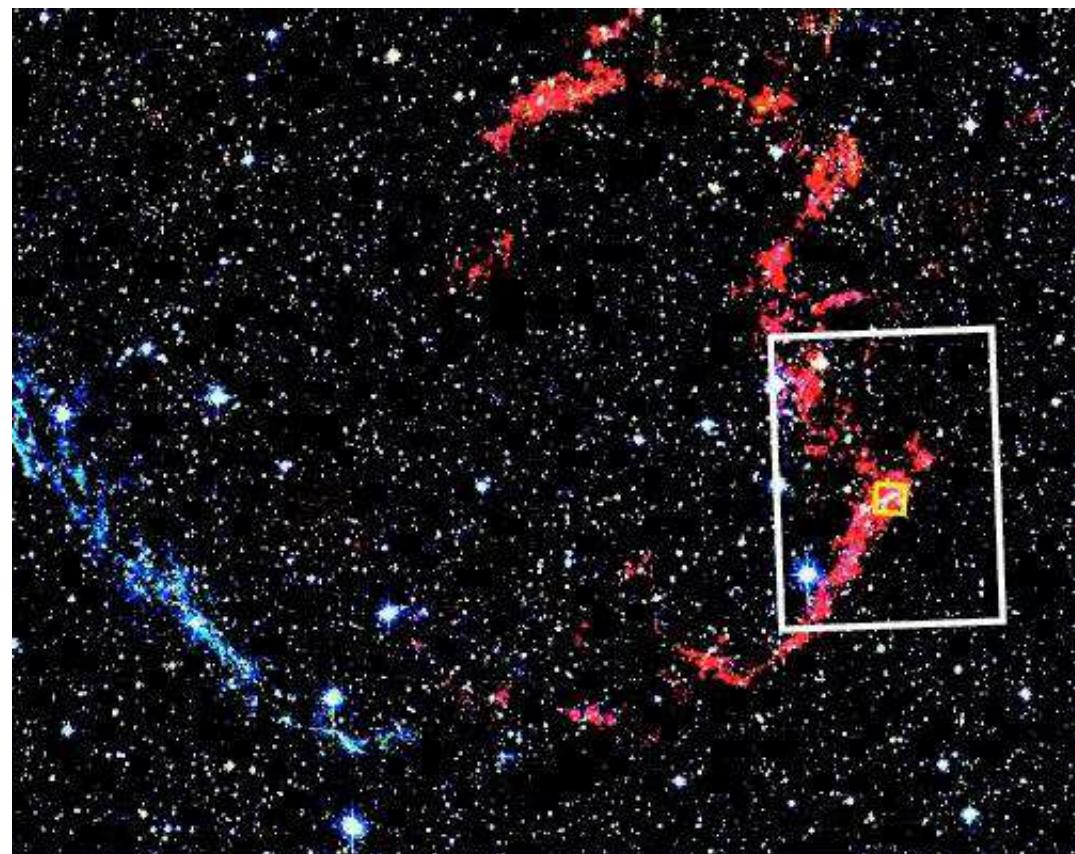
[H₂ Dissociation](#)

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[H₂ and metallicity](#)

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[Conclusion](#)



Neufeld & Yuan, ApJ, 2008 - S443 SNR
Red: K band H₂ 1 – 0 $S(1)$



H₂ Structure

H₂ Structure

H₂ Dissociation

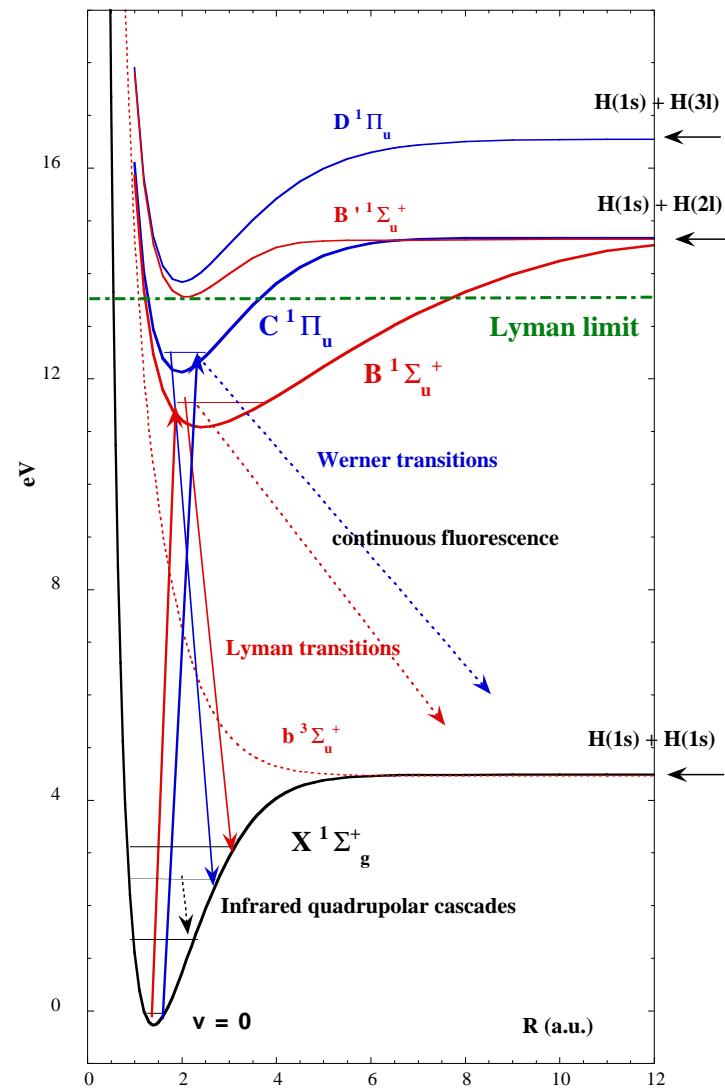
H₂ Formation

H₂ and metallicity

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Conclusion

Absorption and fluorescence in astrophysical media



Homonuclear and light \Rightarrow



H₂ Structure

H₂ Structure

H₂ Dissociation

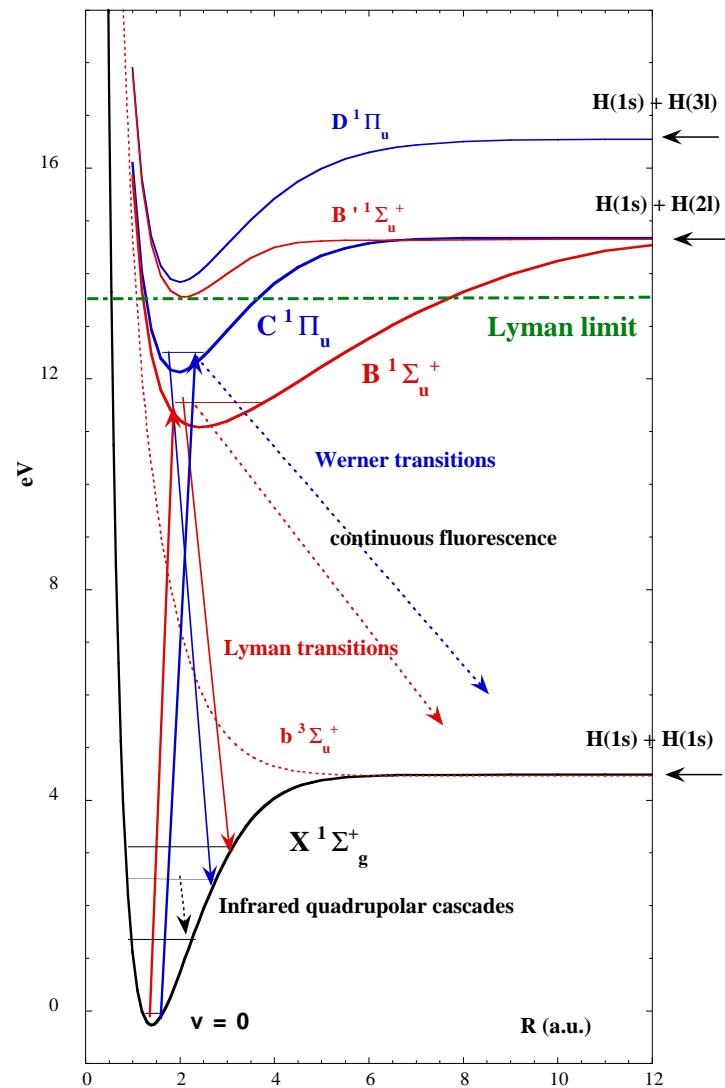
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Absorption and fluorescence in astrophysical media



Homonuclear and light \Rightarrow

■ No dipolar transitions



H₂ Structure

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H₂ Dissociation

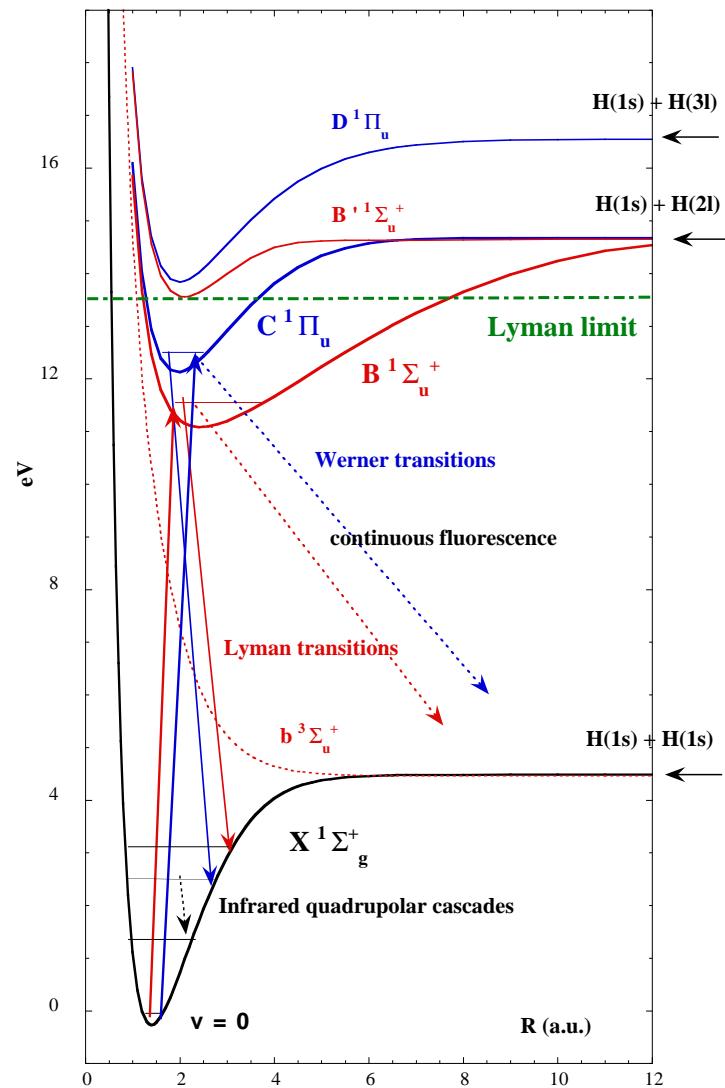
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Homonuclear and light \Rightarrow

- No dipolar transitions
- No formation in gas phase



H₂ Structure

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H₂ Dissociation

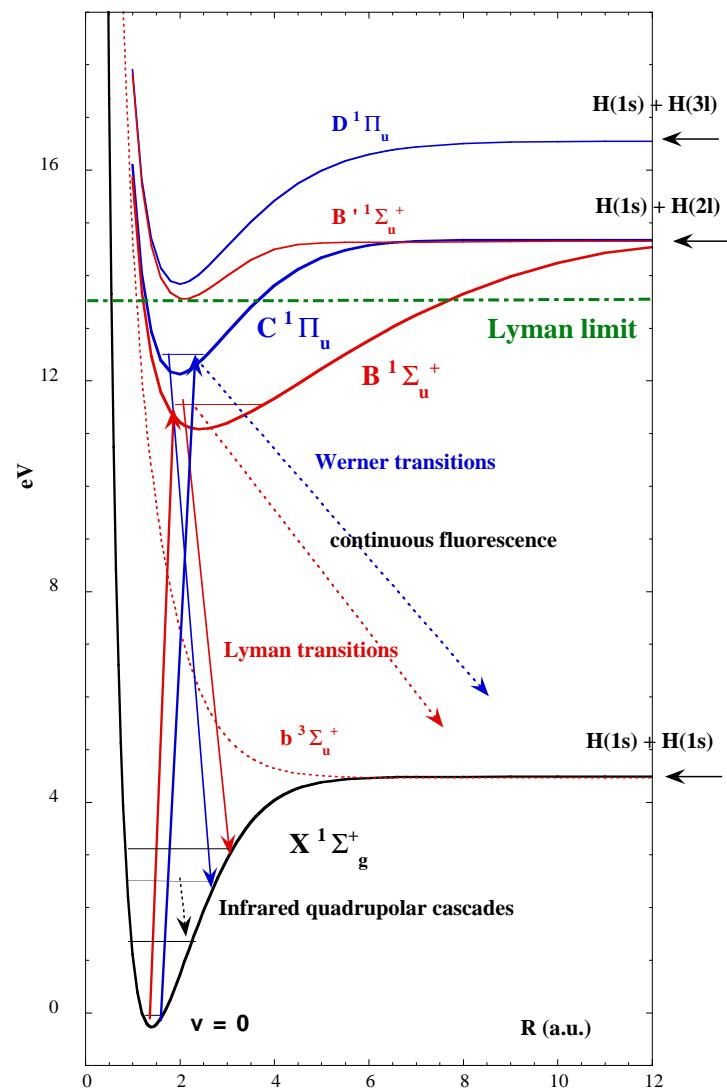
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Absorption and fluorescence in astrophysical media



Homonuclear and light \Rightarrow

- No dipolar transitions
- No formation in gas phase
- No direct UV dissociation



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H₂ Dissociation

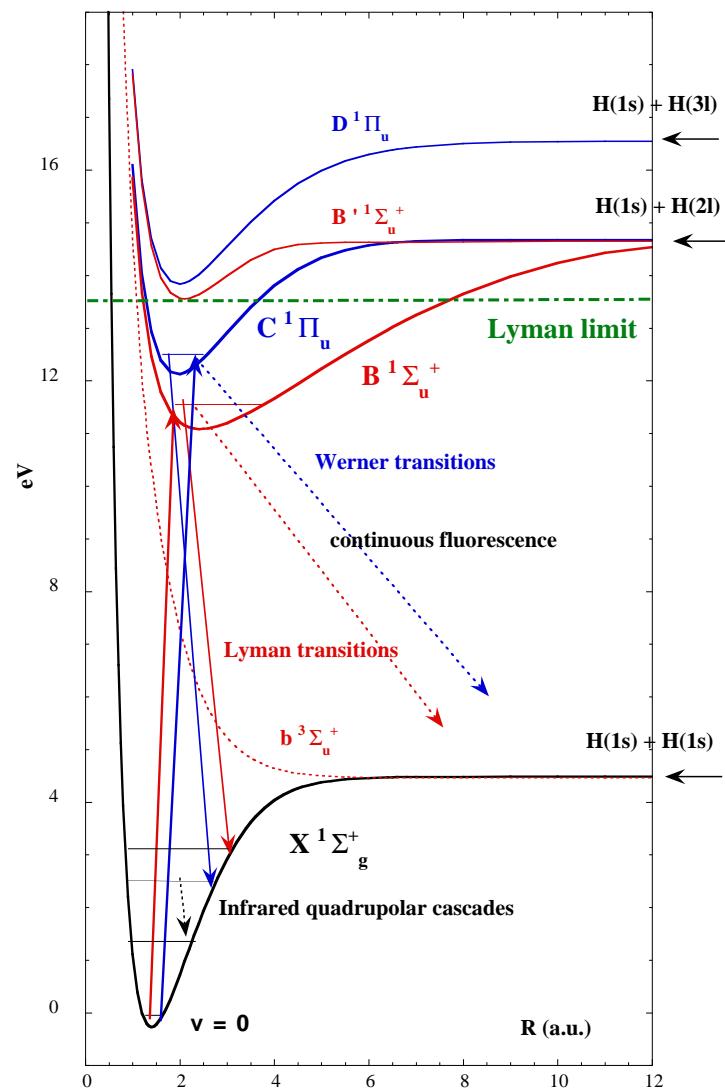
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Homonuclear and light ⇒

- No dipolar transitions
- No formation in gas phase
- No direct UV dissociation
- Formation releases 4.5 eV



H₂ Structure

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H₂ Dissociation

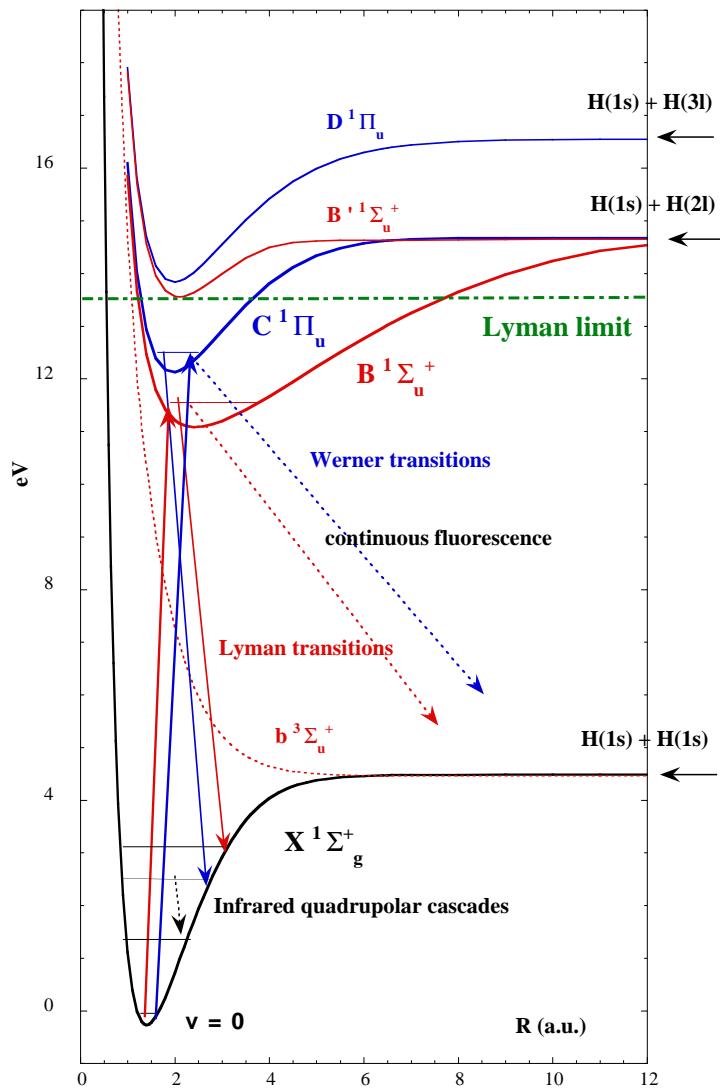
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Absorption and fluorescence in astrophysical media



Homonuclear and light ⇒

- No dipolar transitions
- No formation in gas phase
- No direct UV dissociation
- Formation releases 4.5 eV
- Lowest transitions are high:
 $J = 2 \rightarrow 0 : 509.8 \text{ K}$
 $J = 3 \rightarrow 1 : 844.6 \text{ K}$
- Source of secondary photons.



H₂ Destruction

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■ UV transitions:

- ◆ Pumping in Lyman ($B^1\Sigma_u^+$) and Werner systems ($C^1\Pi_u$)
- ◆ Discrete transition, either to $X^1\Sigma_g^+$ (88 %) or to the continuum.
- ◆ In free space, for Draine's field $k_D^0 = 5.8 \cdot 10^{-11} \text{ s}^{-1}$.



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■ Within a cloud:

$$k_D = k_D^0 I_{UV} \exp(-\sigma_g (N_1 + 2 N_2)) f_{shield}(N_2)$$

N_1 & N_2 : Column densities of H and H₂,

σ_g : grain absorption, I_{UV} : radiation field strength.



H₂ Self-Shielding

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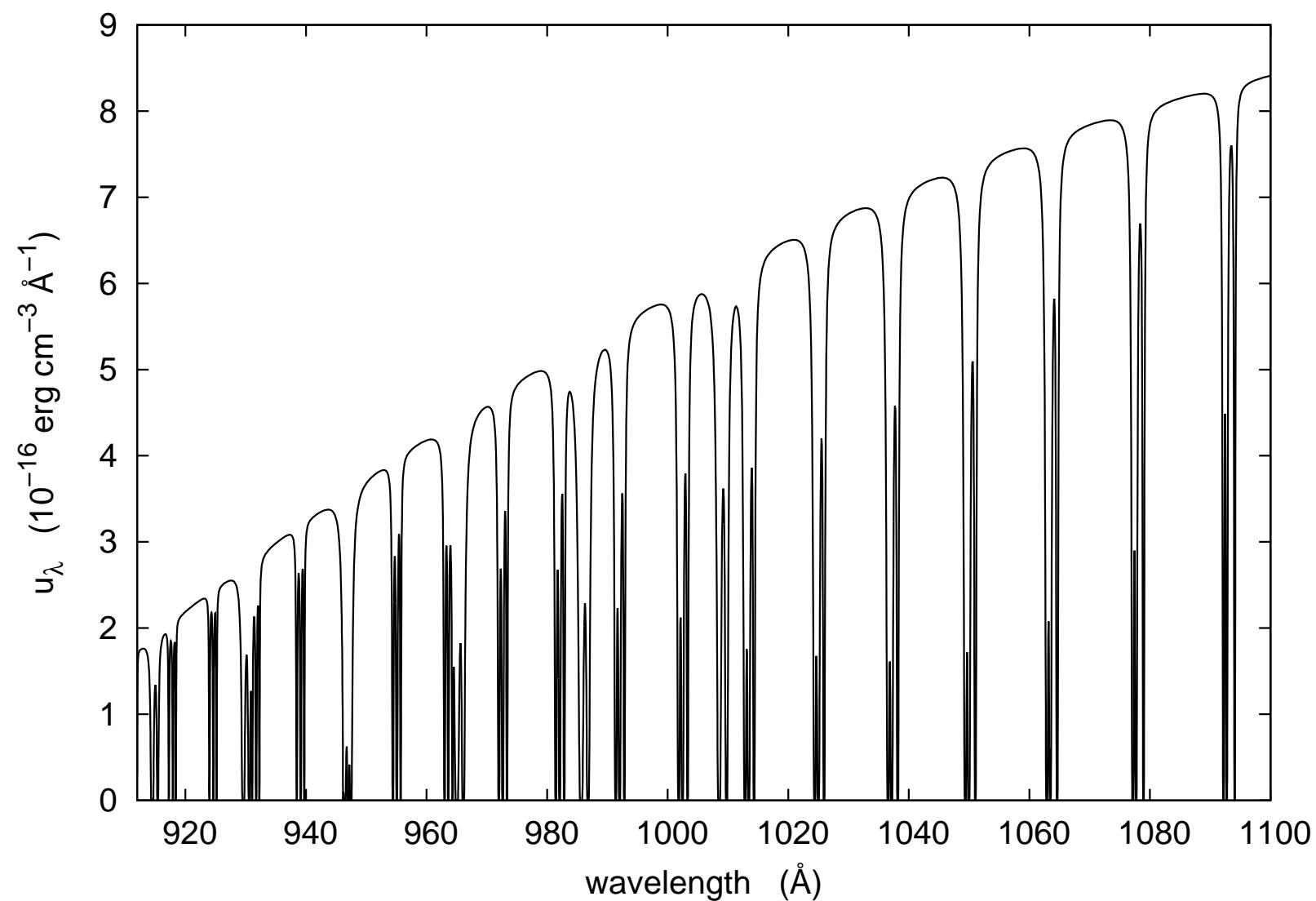
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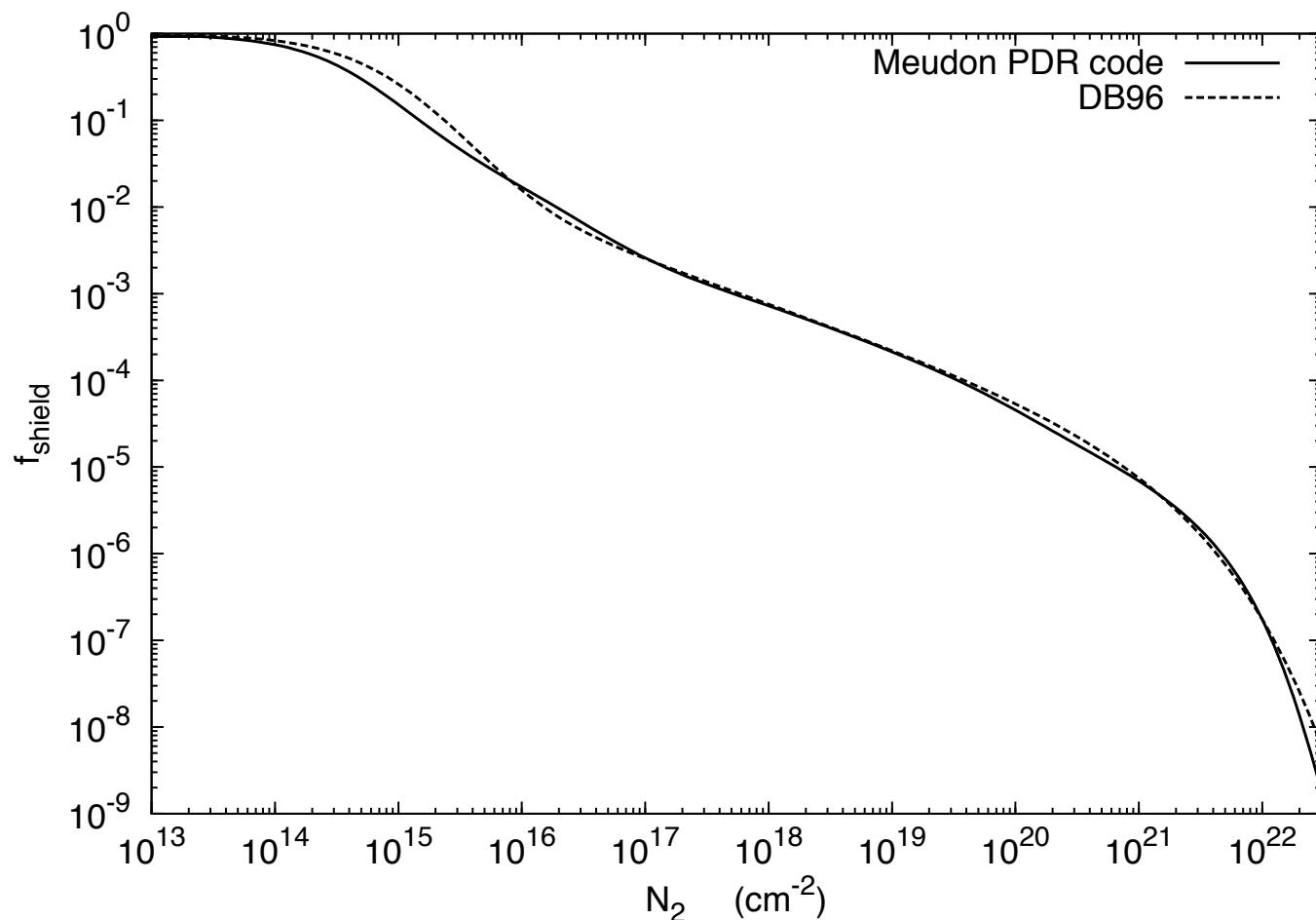
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Self-shielding function: $f_{\text{shield}} = \frac{1}{\sigma_d} \frac{dW_d}{dN_2}$. $\sigma_d \simeq 2.36 \cdot 10^{-3} \text{ cm}^2 \text{ Hz}$



H₂ Formation

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- Formation on grains \Rightarrow by default

$$\frac{d[\text{H}_2]}{dt} = R_f n_{\text{H}} n(\text{H})$$

- Copernicus (Jura, (1975) ApJ 197, 575):

$$R_f \sim 3 \cdot 10^{-17} \text{ cm}^3 \text{ s}^{-1}$$

- Details still under investigation
(see talk by Emeric Bron, following mine)



H_2 formation(traditional view)

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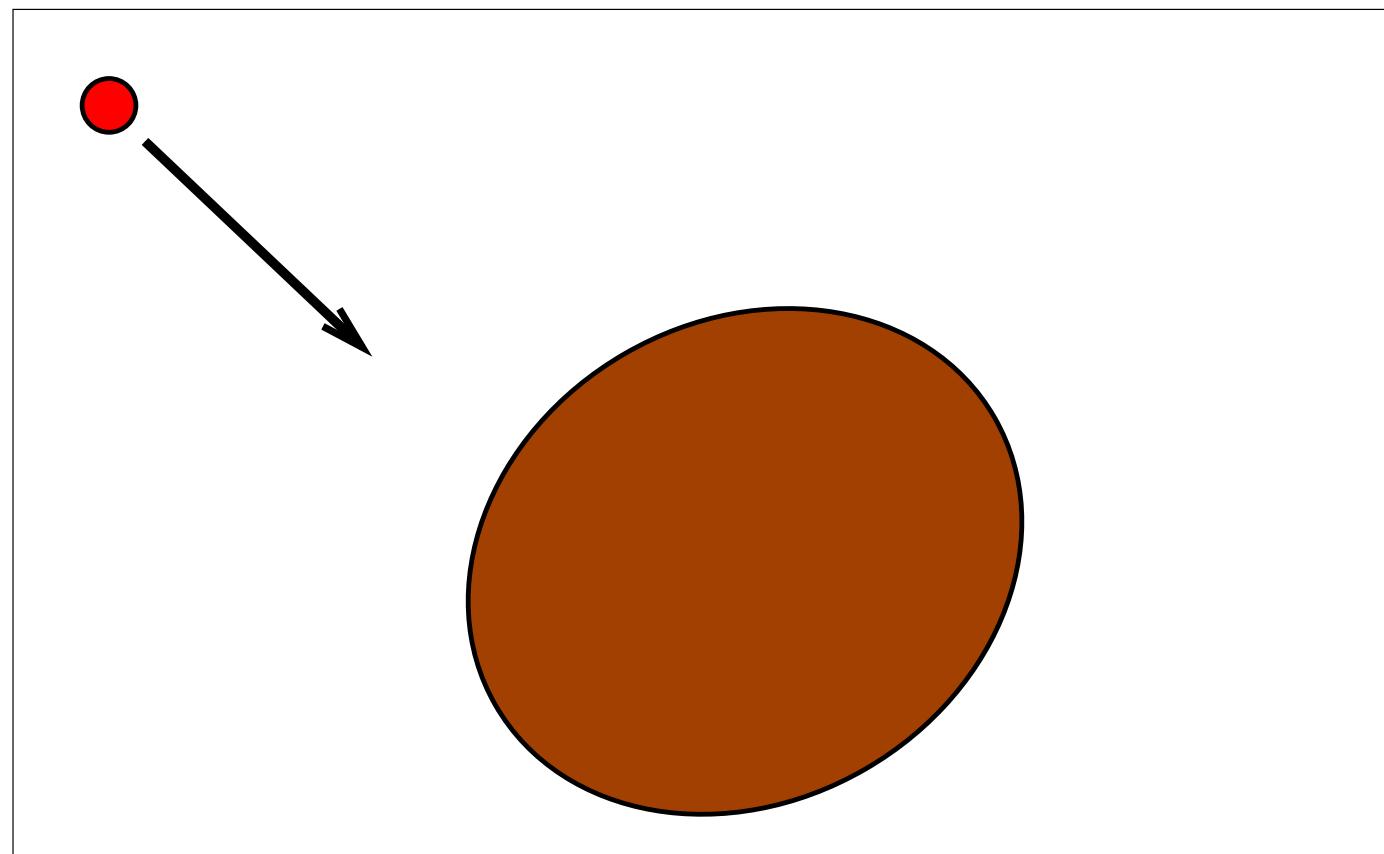
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- Rate: ?



H_2 formation (traditional view)

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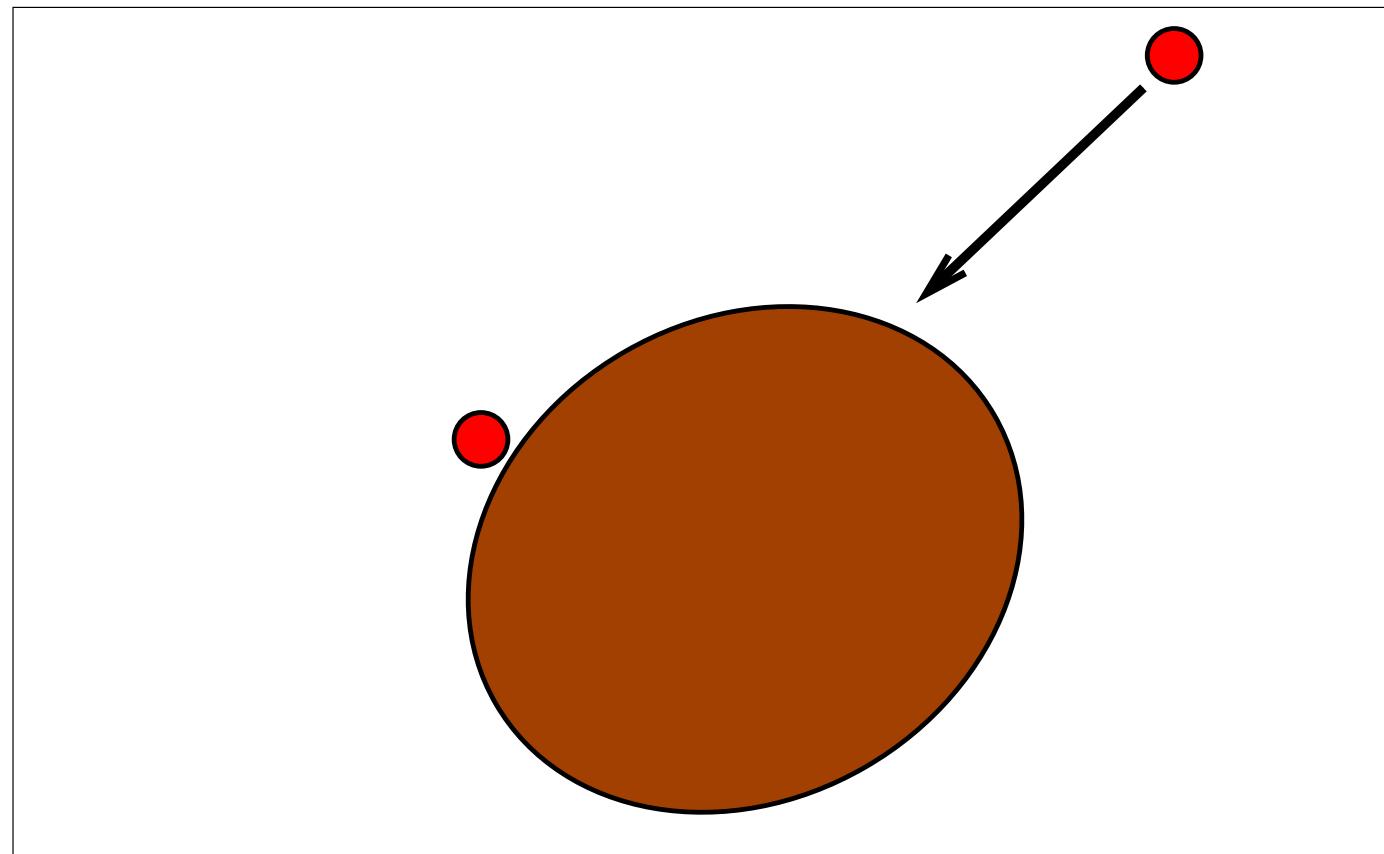
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- Rate: known from observations



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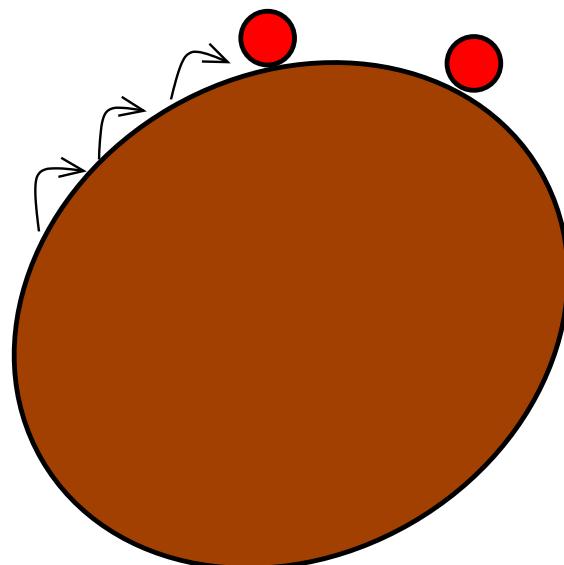
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- Rate: Proportional to v_T , $n(H)$ and n_{gr}



H₂ formation (traditional view)

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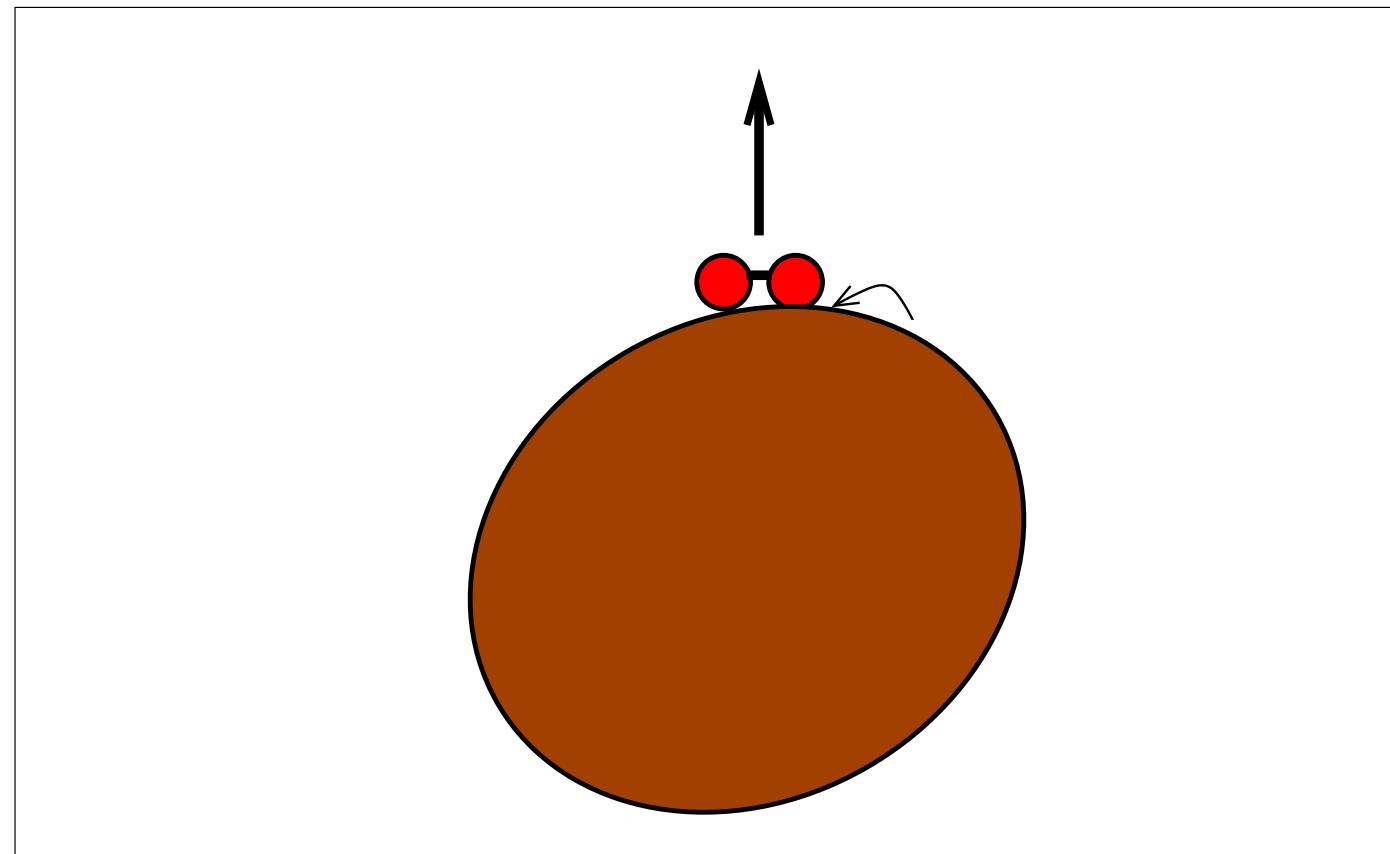
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- Rate: $\sim 3 \cdot 10^{-17} \sqrt{\frac{T}{100}} Z' n_H n(\text{H}) \text{ cm}^{-3} \text{ s}^{-1}$
- Langmuir-Hinshelwood (LH)



H₂ formation (traditional view)

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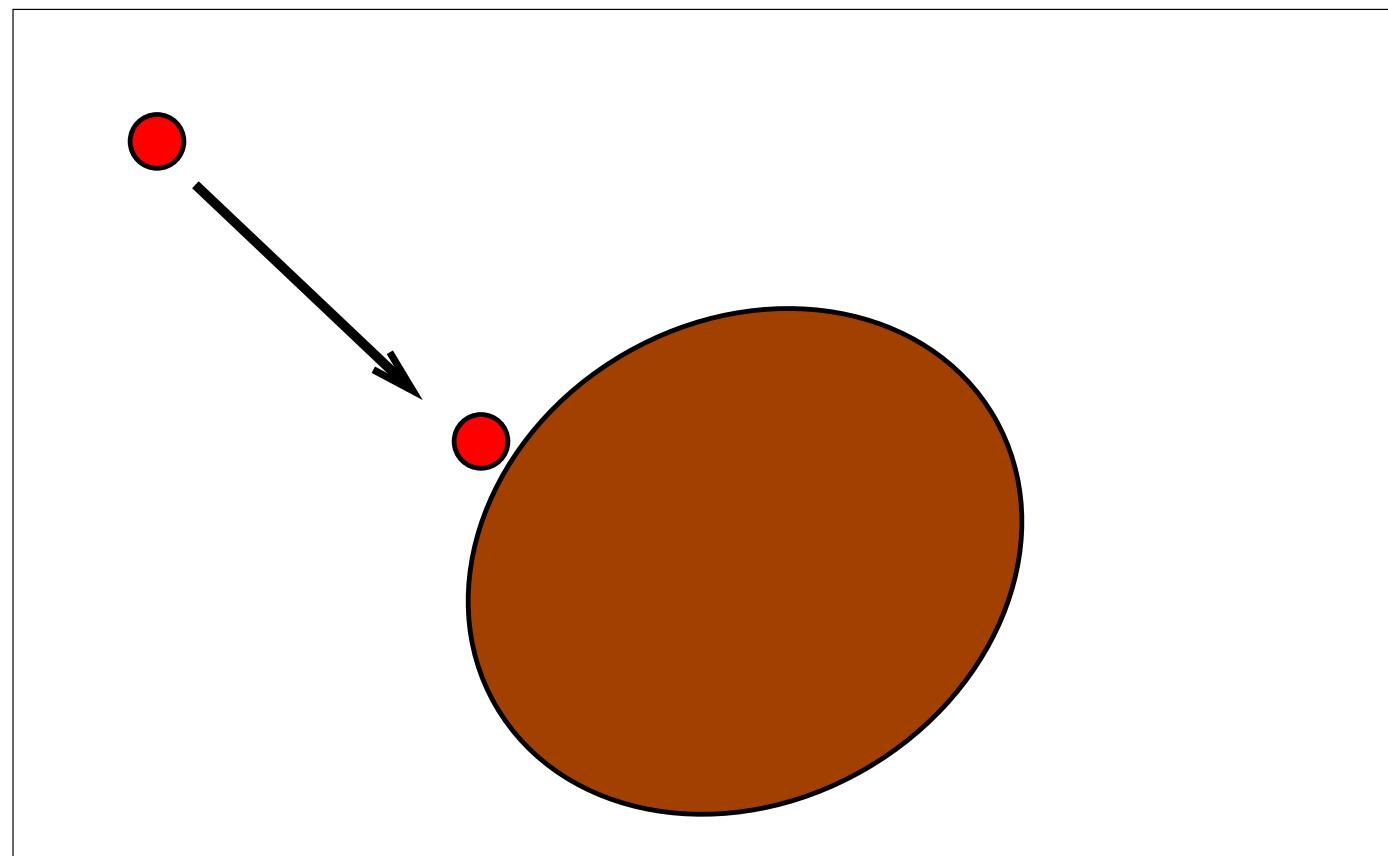
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- Rate: $\sim 3 \cdot 10^{-17} \sqrt{\frac{T}{100}} Z' n_H n(\text{H}) \text{ cm}^{-3} \text{ s}^{-1}$
- Eley-Rideal (ER)



H₂ formation - Full computation

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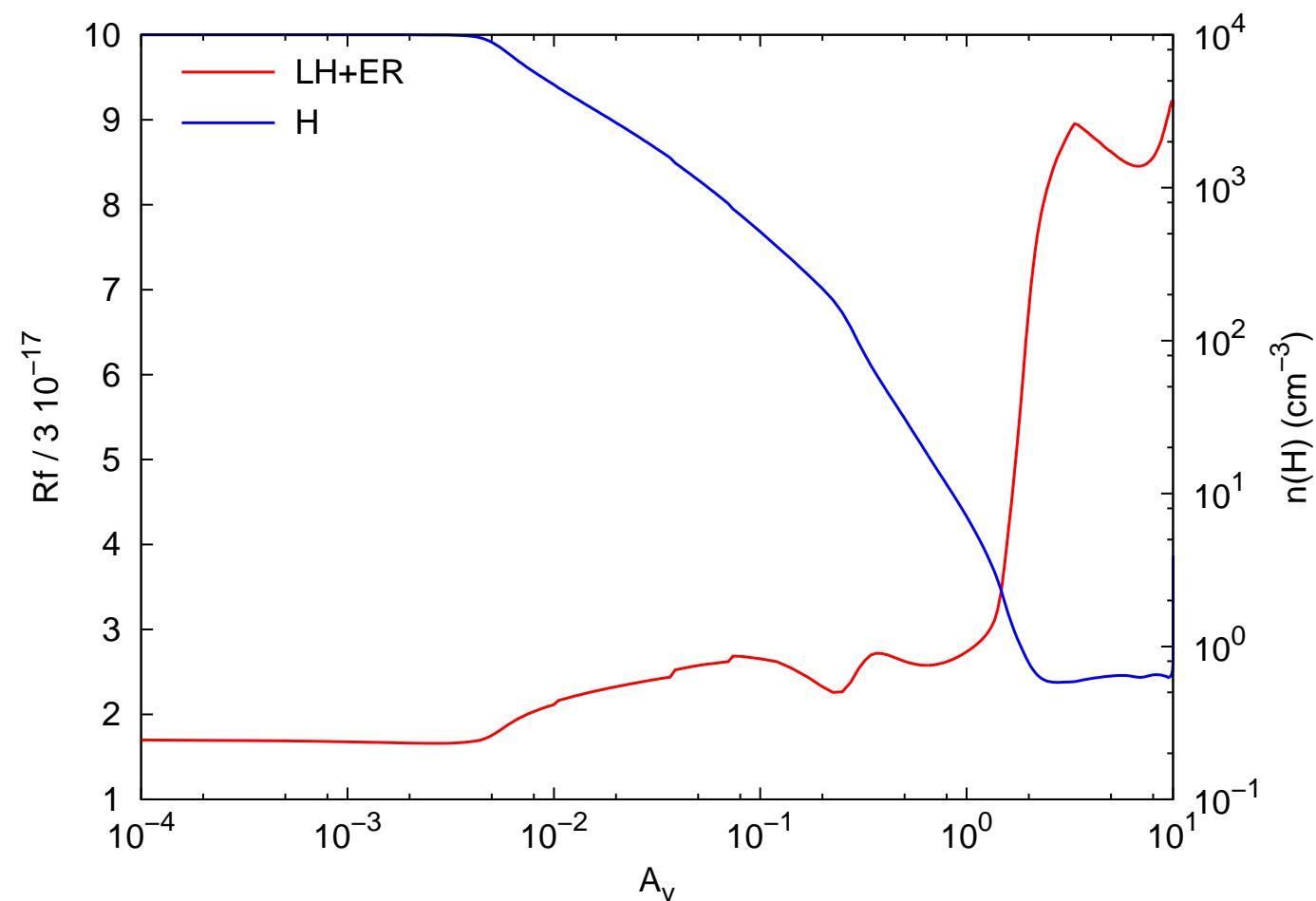
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- $n_{\text{H}} = 10^4 \text{ cm}^{-3}$, Mathis $\times 100$





H₂ formation - Problem I

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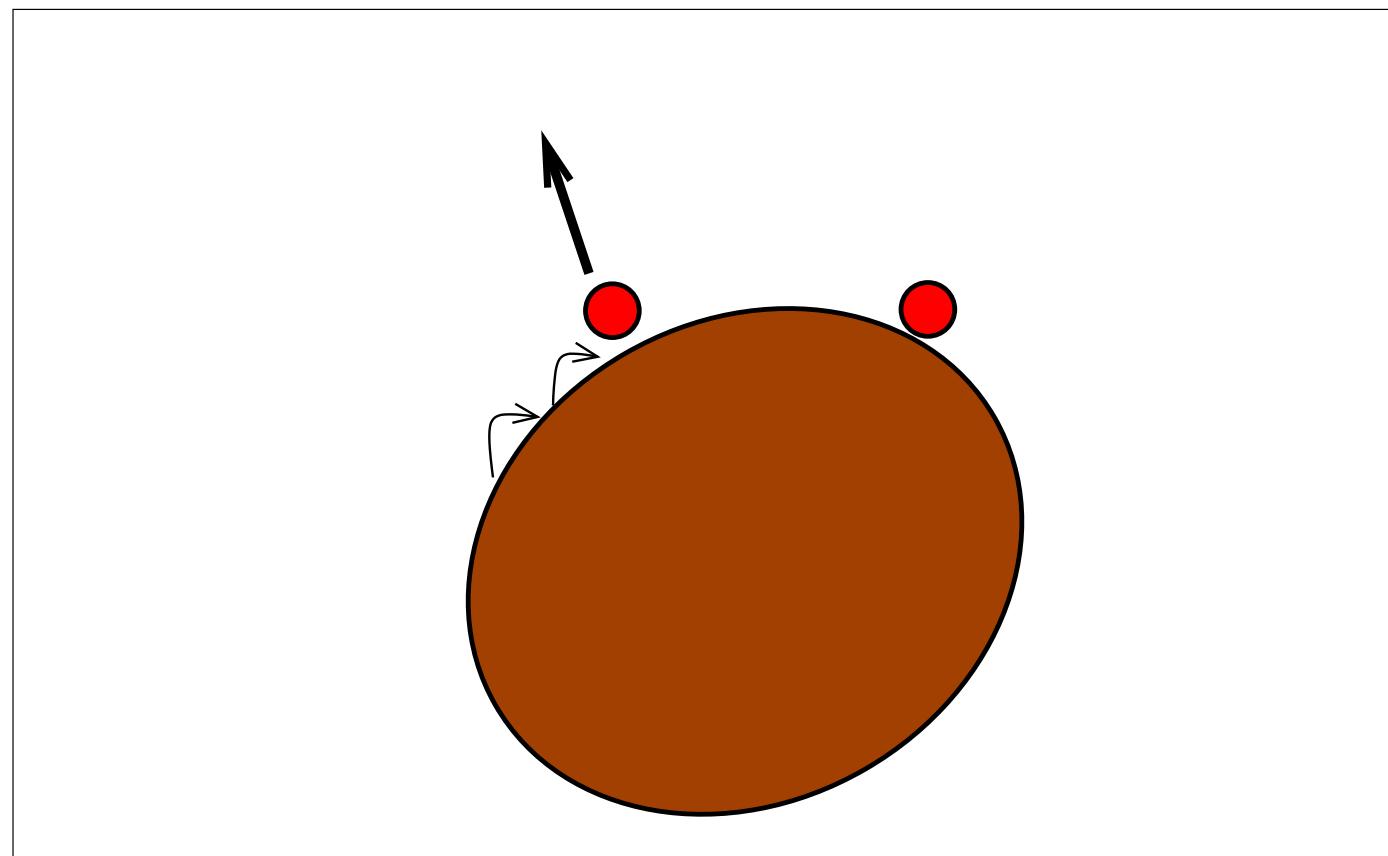
Problems

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- Evaporation is fast from hot grains.



H₂ formation - Problem II

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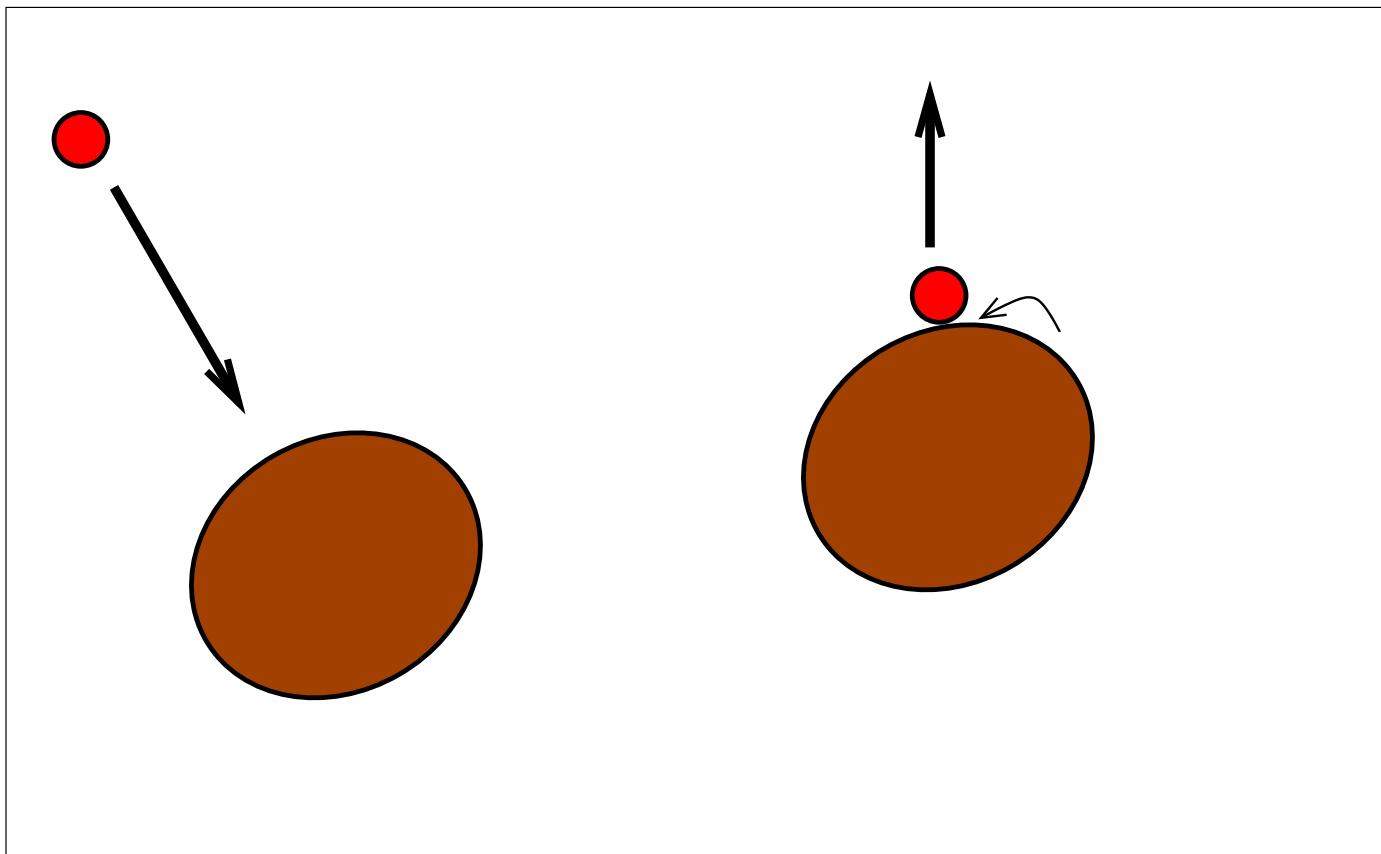
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- Grains are **very** small.



H_2 formation - Fluctuations rescue!

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- Requires statistical computation of fluctuations
- See following presentation by Emeric Bron
Bron et al. (2014), A&A 569, 100



H_2 formation - Fluctuations rescue!

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- Requires statistical computation of fluctuations
- See following presentation by Emeric Bron
Bron et al. (2014), A&A 569, 100
- Take home message:
 - ◆ Crude evaluation: rate equations OK
(e.g. in MHD codes).
 - ◆ Grain physics: full statistical formalism required.



H/H₂ transition

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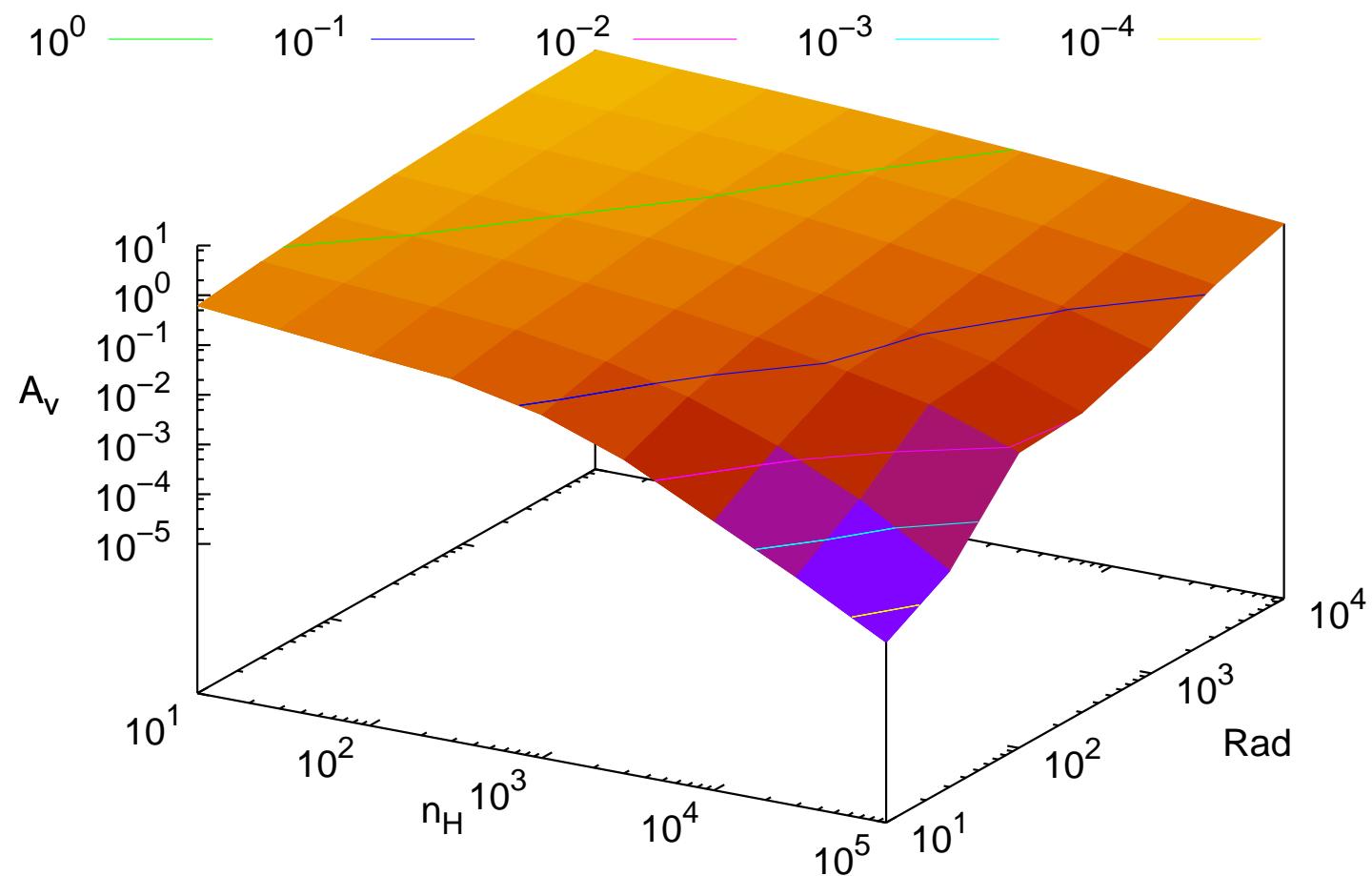
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- To first order, depends on I_{UV}/n_H :





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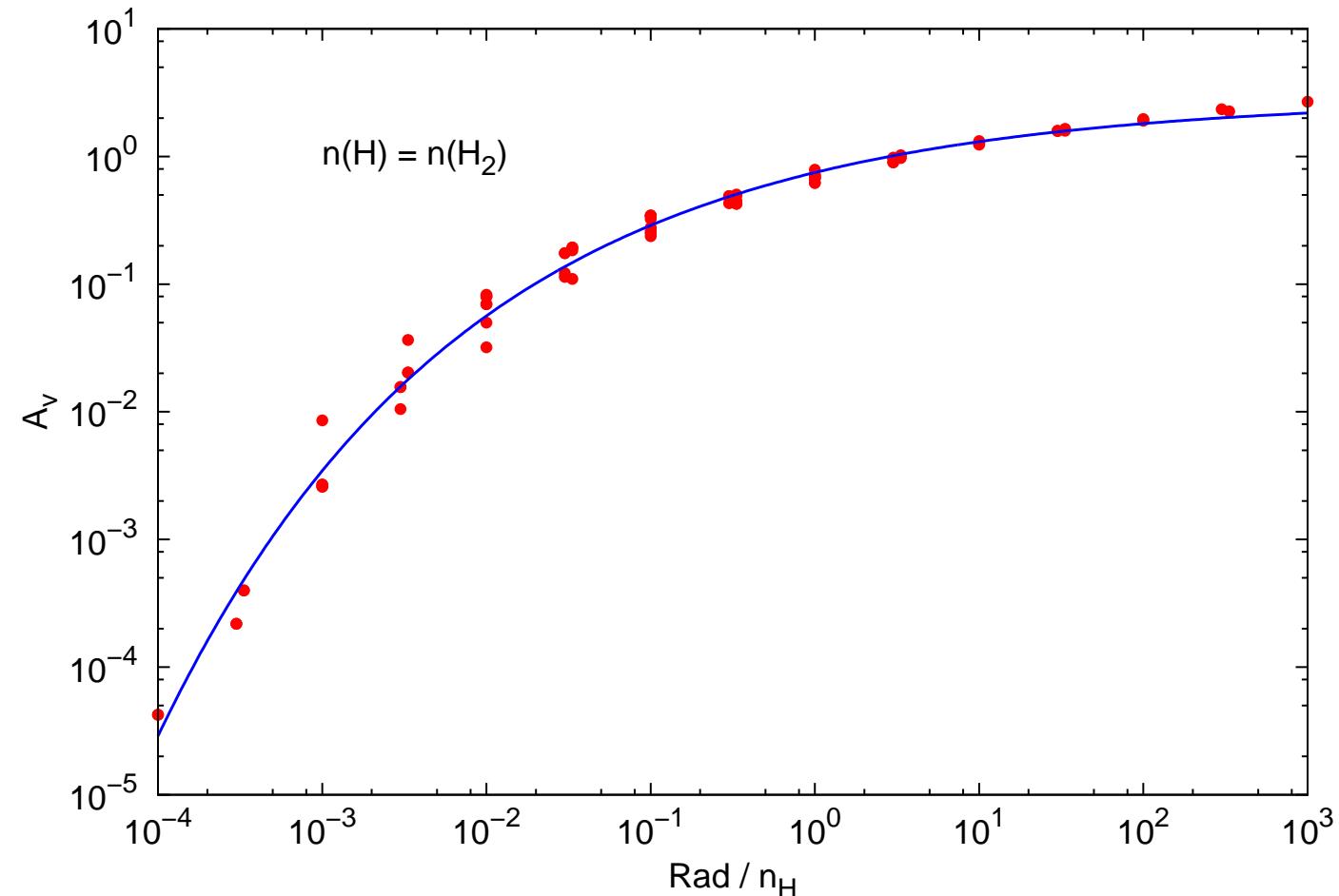
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H₂ and metallicity

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N(H) from N(H₂)

Analysis

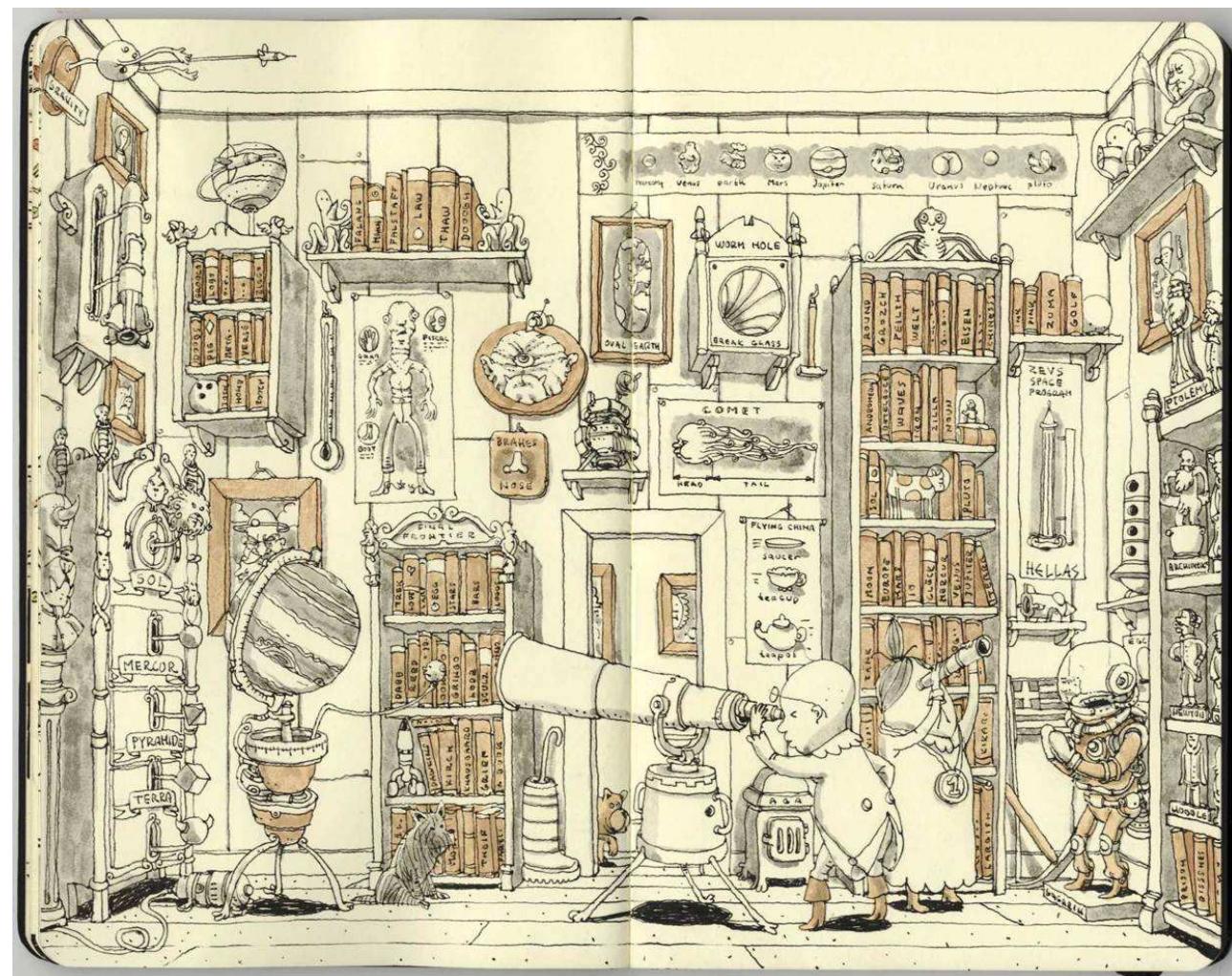
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Adapted from Sternberg et al. (2014), ApJ 790, 10.



www.behance.net/MattiasA



H₂ Constant Formation/Destruction Approximation

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- With a constant formation rate R_f and density n_{H} :

$$R_f n_{\text{H}} n(\text{H}) = k_D n(\text{H}_2)$$



H₂ Constant Formation/Destruction Approximation

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- With a constant formation rate R_f and density n_{H} :

$$R_f n_{\text{H}} n(\text{H}) = k_D n(\text{H}_2)$$

$$R_f = 3 \cdot 10^{-17} \sqrt{\frac{T}{100}} Z' \text{ cm}^3 \text{ s}^{-1}$$

$$k_D = k_D^0 f(N_2) \exp(-\sigma_g (N_1 + 2N_2)) \text{ s}^{-1}$$

$$\sigma_g \simeq 1.9 \cdot 10^{-21} Z' \text{ cm}^2 \text{ H}^{-1}$$



H₂ Constant Formation/Destruction Approximation

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$$\sigma_g \simeq 1.9 \cdot 10^{-21} Z' \text{ cm}^2 \text{ H}^{-1}$$

- Equation is separable:

$$n(\text{H}) = \frac{dN(\text{H})}{ds} = \frac{dN_1}{ds}; \quad n(\text{H}_2) = \frac{dN(\text{H}_2)}{ds} = \frac{dN_2}{ds}$$

$$R_f n_{\text{H}} \exp(\sigma_g N_1) dN_1 = k_D^0 f(N_2) \exp(-\sigma_g 2N_2) dN_2$$



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$$N_1 = \frac{1}{\sigma_g} \log \left(1 + \frac{k_D^0 \sigma_g}{R_f n_{\text{H}}} \int_0^{N_2} f(N) \exp(-2\sigma_g N) dN \right)$$



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$$N_1 = \frac{1}{\sigma_g} \log \left(1 + \frac{k_D^0 \sigma_g}{R_f n_{\text{H}}} \int_0^{N_2} f(N) \exp(-2\sigma_g N) dN \right)$$

Draine and Bertoldi (1996) give:

$$f(x) = \frac{0.965}{\left(1 + \frac{x}{b_5}\right)^2} + \frac{0.035}{(1+x)^{0.5}} \exp\left(-8.5 \cdot 10^{-4} (1+x)^{0.5}\right)$$

with $x = \frac{N}{N_0}$, $N_0 = 5 \cdot 10^{14} \text{ cm}^{-2}$ and $b_5 = \frac{b}{10^5 \text{ cm s}^{-1}}$
⇒ Analytical integration is possible.



Comparison to model

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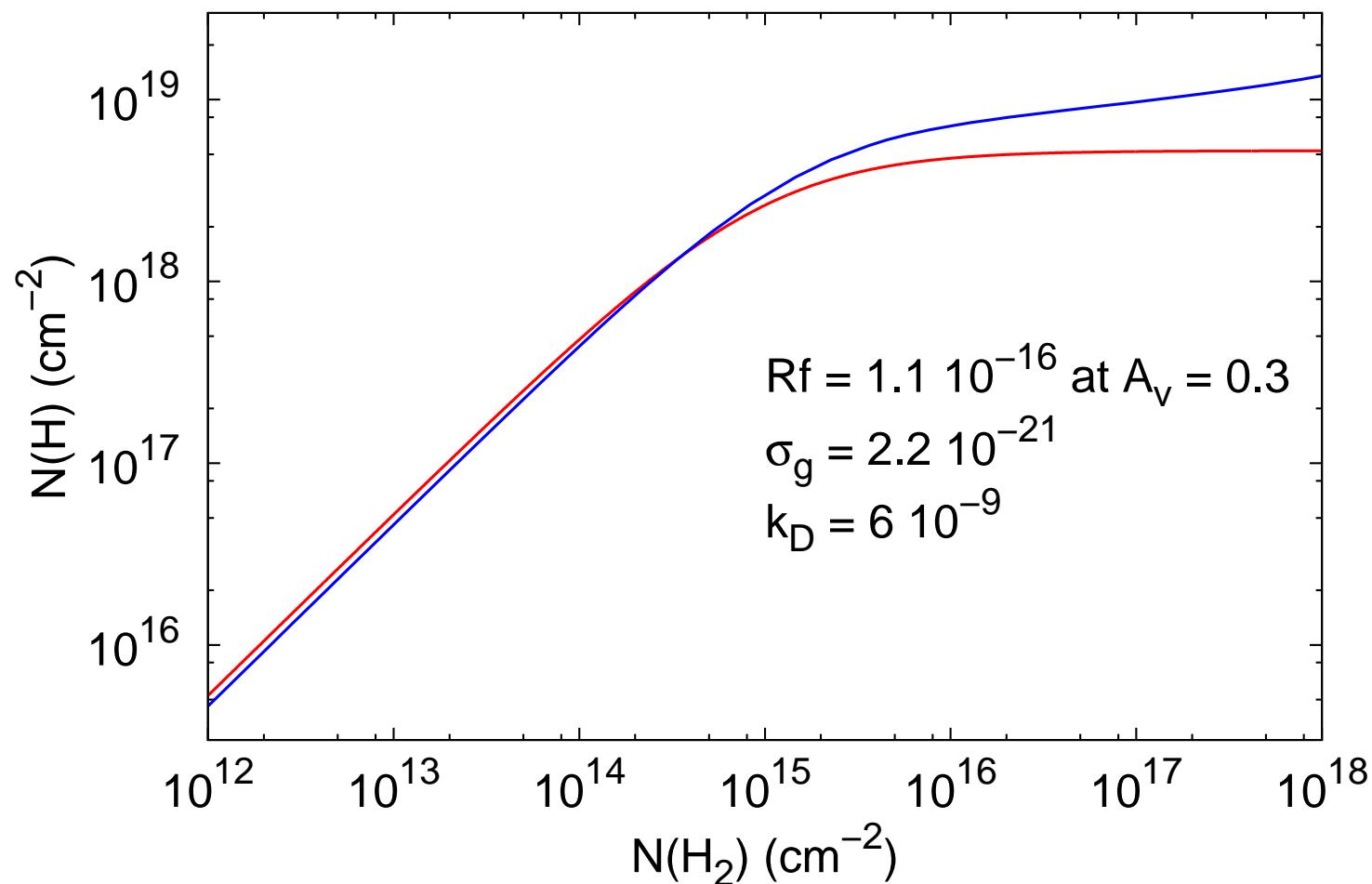
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Notice effect of chemistry at high $N(H_2)$.



Analytical analysis

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Back to:

$$N_1(N_2) = \frac{1}{\sigma_g} \log \left(1 + \frac{k_D^0 \sigma_g}{R_f n_{\text{H}}} \int_0^{N_2} f(N) \exp(-2\sigma_g N) dN \right)$$



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Back to:

$$N_1(N_2) = \frac{1}{\sigma_g} \log \left(1 + \frac{k_D^0 \sigma_g}{R_f n_{\text{H}}} \int_0^{N_2} f(N) \exp(-2\sigma_g N) dN \right)$$

We set:

$$\alpha = \frac{2 k_D^0}{R_f n_{\text{H}}}; \quad G(N_2) = \sigma_g \int_0^{N_2} f(N) \exp(-2\sigma_g N) dN$$

$$N_1(N_2) = \frac{1}{\sigma_g} \log \left(1 + \frac{\alpha G(N_2)}{2} \right)$$



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Back to:

$$N_1(N_2) = \frac{1}{\sigma_g} \log \left(1 + \frac{k_D^0 \sigma_g}{R_f n_{\text{H}}} \int_0^{N_2} f(N) \exp(-2\sigma_g N) dN \right)$$

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$$N_1(N_2) = \frac{1}{\sigma_g} \log \left(1 + \frac{\alpha G(N_2)}{2} \right)$$

- $\alpha = \frac{n_1}{n_2}$ in free space (typically: $\alpha \sim 2 \cdot 10^4$)
- G : Average H₂ self-shielding factor. (typically $G \sim 5 \cdot 10^{-5}$)



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- Asymptotically:

$$N_{1,tot} = \frac{1}{\sigma_g} \log \left(1 + \frac{\alpha G}{2} \right); \quad G = \lim_{N_2 \rightarrow \infty} G(N_2)$$

- Most Z' dependancies cancel, but one:

$$\alpha G \simeq 1.5 \frac{I_{UV}}{(n_H/100 \text{ cm}^{-3})} \frac{1}{1 + \sqrt{2.64 Z'}}$$



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- $\alpha G \ll 1$: weak field limit:
Absorption by H₂ lines and H₂-dust dominates.
- $\alpha G \gg 1$: strong field limit:
Absorption by HI-dust dominates.

See complete discussion in Sternberg et al. (2014)



Weak to strong field

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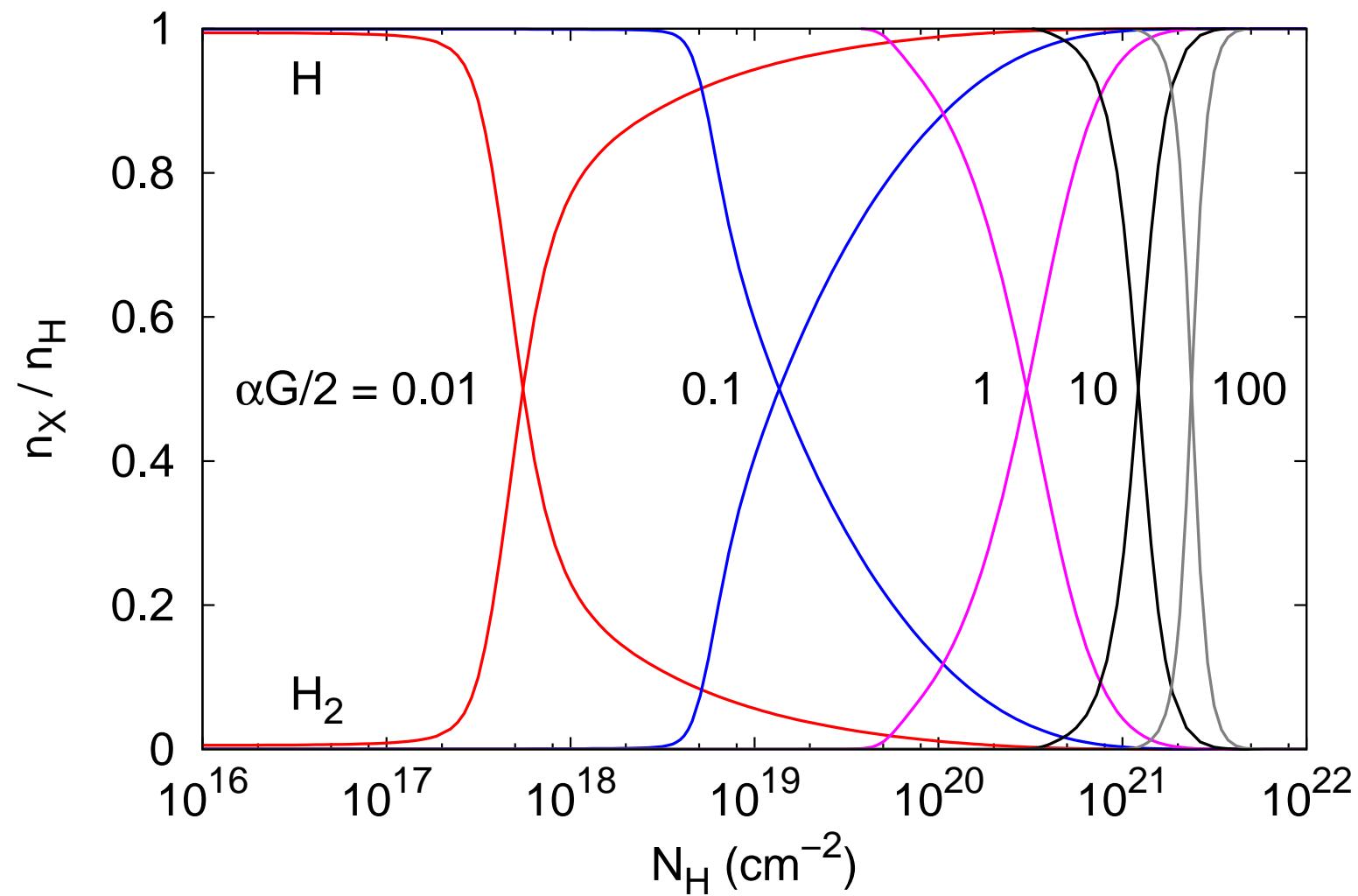
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Effects of metallicity

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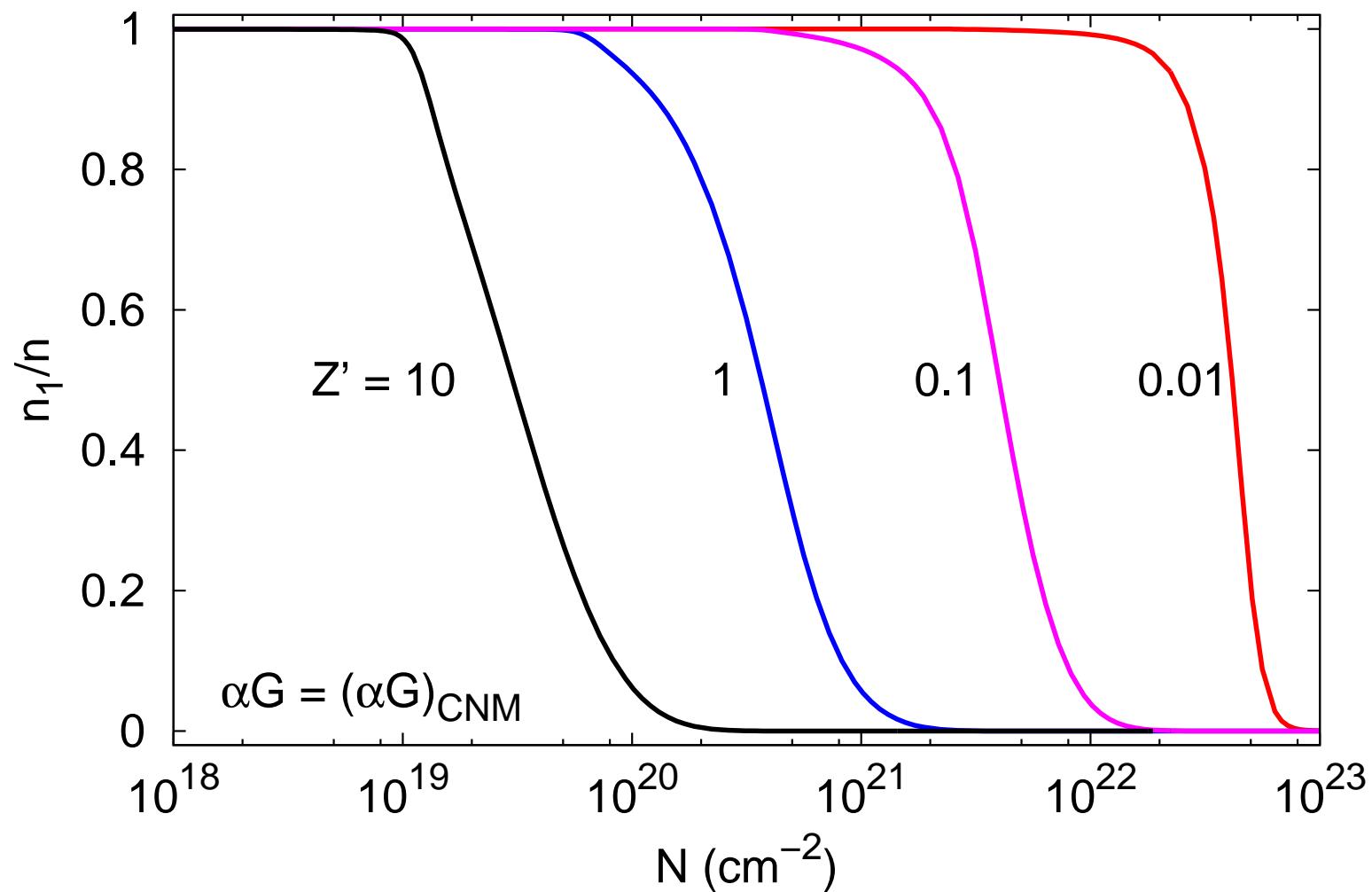
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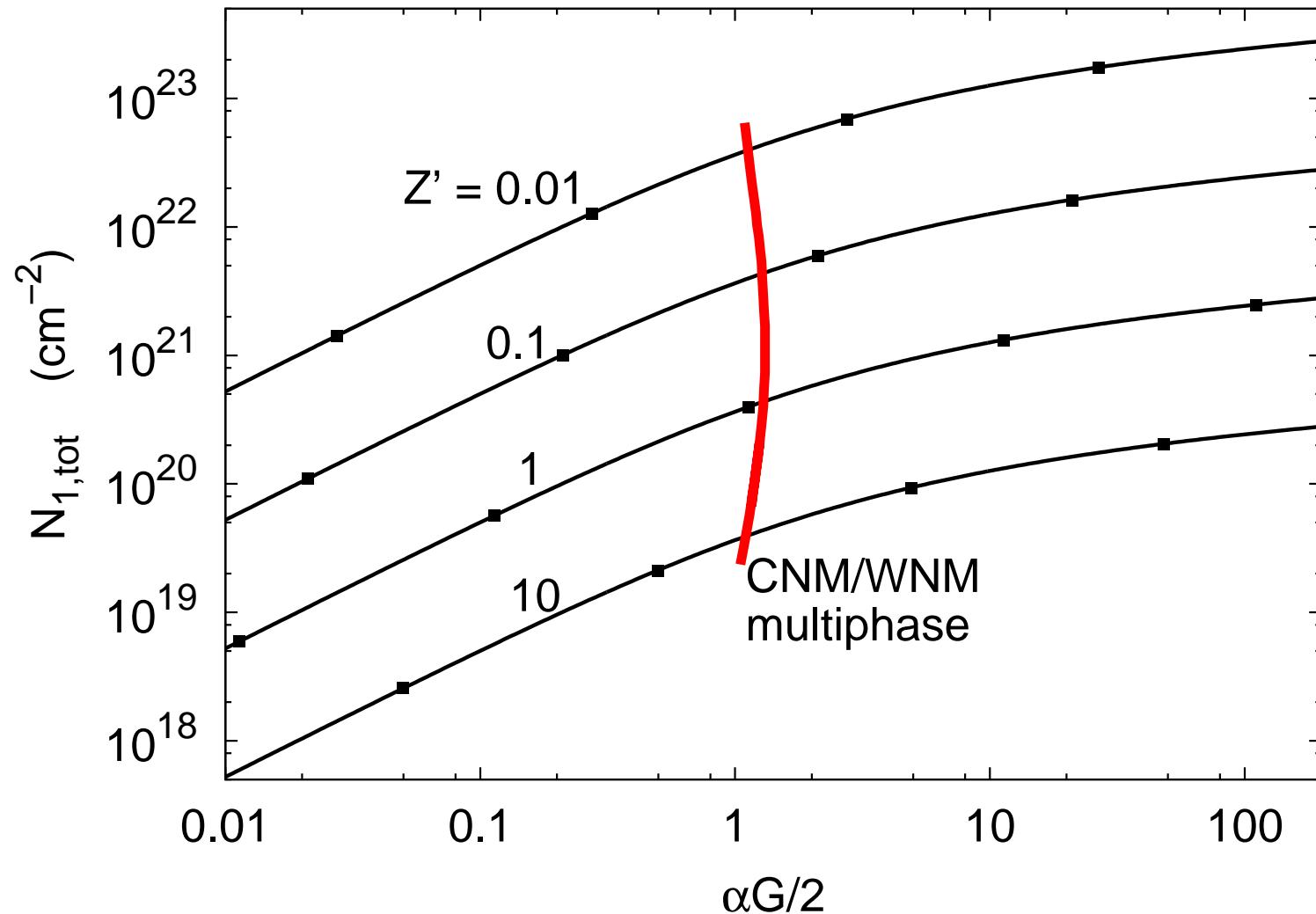
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For isotropic radiation field:

$$N_{1,tot} = \frac{\langle \mu \rangle}{\sigma_g} \log \left(1 + \frac{1}{\langle \mu \rangle} \frac{\alpha G}{4} \right)$$

- $\langle \mu \rangle \simeq 0.8$, from fit to numerical models
- $\frac{1}{\sigma_g} = \frac{5.3 \cdot 10^{20}}{Z'} \text{ cm}^{-2}$, for typical grain composition
- $\alpha G \simeq 1.5 \frac{I_{UV}}{(n_H/100 \text{ cm}^{-3})} \frac{1}{1 + \sqrt{2.64 Z'}}$
- If balance between WNM and CNM, then $\frac{\alpha G}{2} \simeq 1.1$, so:

$$N_{1,tot} \simeq \frac{2.2 \cdot 10^{20}}{Z'} \text{ cm}^{-2}$$



Effects of metallicity

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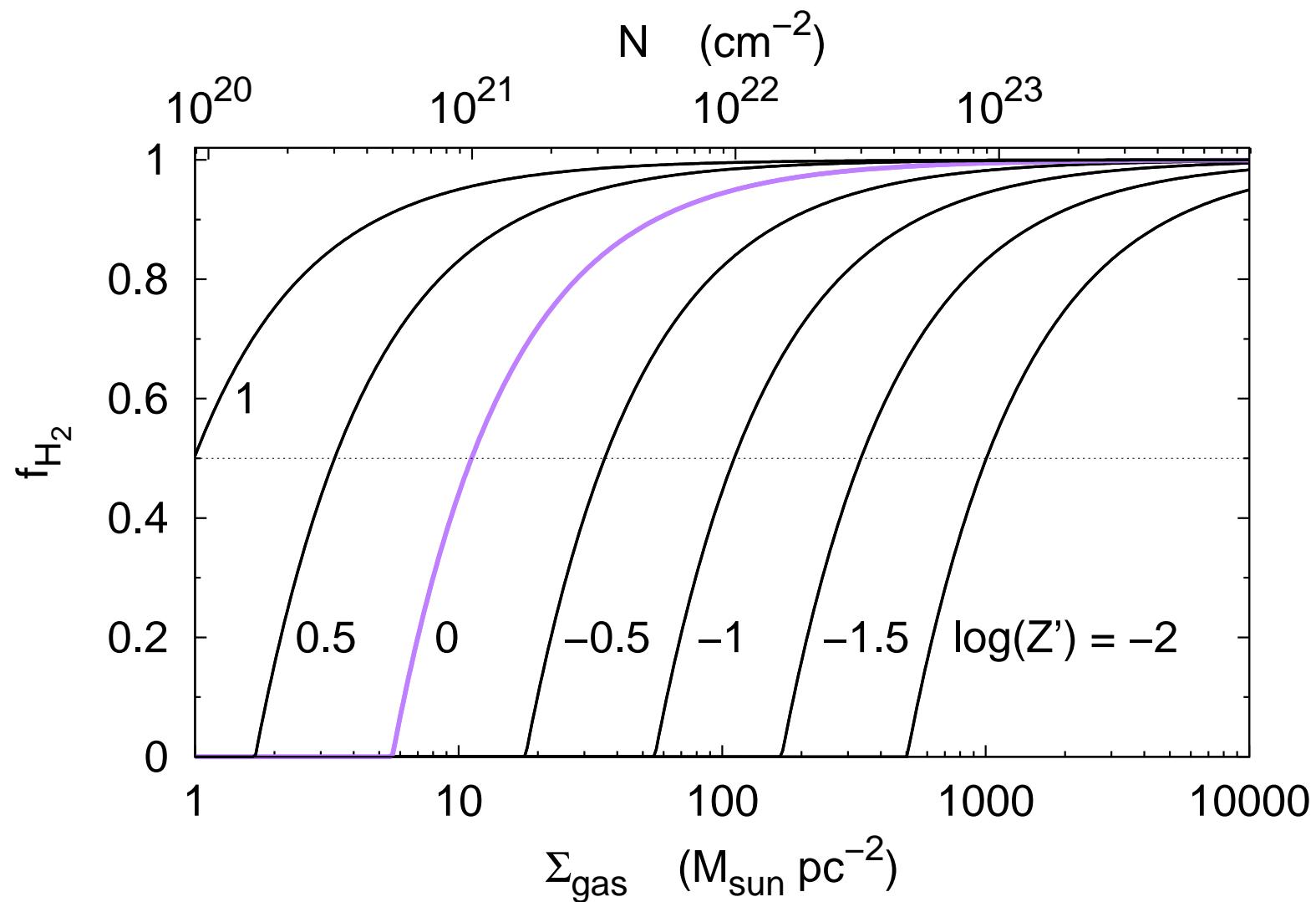
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Expansion

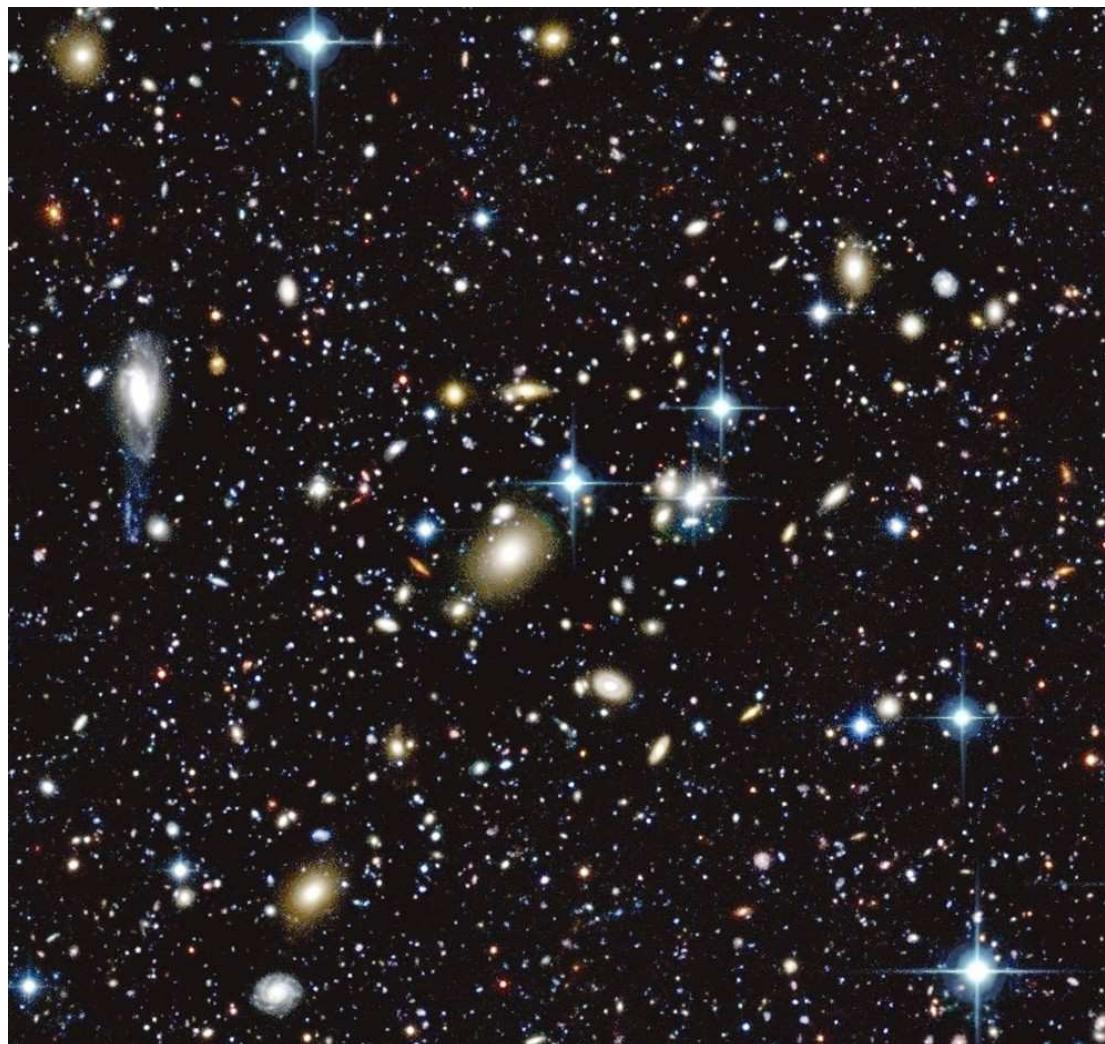
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Expansion

From: “The Dawn of Chemistry”,
D. Galli & F. Palla, ARA&A, 51, 163 (2013)

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From: “The Dawn of Chemistry”,
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- Density (baryons):

$$n_b = 2.2 \cdot 10^{-7} (1+z)^3 \text{ cm}^{-3}$$

- Recombination: $z \sim 1000 \Rightarrow n_b \sim 220 \text{ cm}^{-3}$
- CMB, $T_0 = 2.725 \text{ K}$:

$$T_r = T_0 (1+z)$$

- Gas temperature:

$$\begin{cases} T_g \simeq T_r ; & z > 300 \\ T_g \simeq 0.02 (1+z)^2 ; & z < 100 \end{cases}$$



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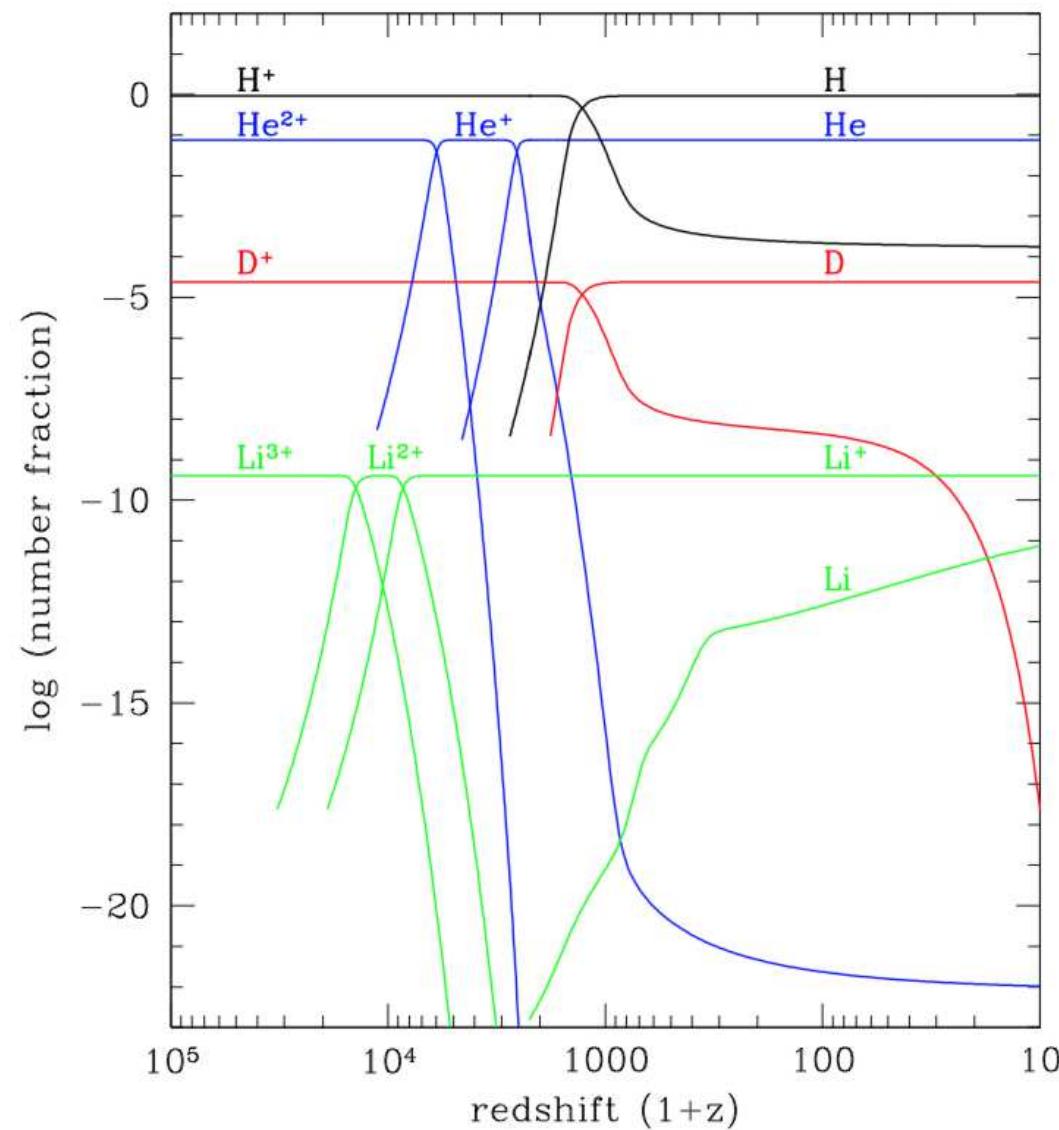
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H	He	D	Li
0.924	0.076	$2.38 \cdot 10^{-5}$	$4.04 \cdot 10^{-10}$

- Full chemistry (with isotopes): ~ 250 reactions and ~ 30 species.



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- Requires detailed balance of all main species (at least vibrational excitation for H₂ and H₂⁺)
- Photo-destruction by CMB and non-thermal photons (e.g. Ly α) is important.
- But also new processes. E.g.

PHYSICAL REVIEW A **85**, 043411 (2012)

Resonances in photoionization: Cross sections for vibrationally excited H₂

J. Zs. Mezei,^{1,2,*} I. F. Schneider,^{1,†} E. Roueff,³ and Ch. Junge^{2,‡}



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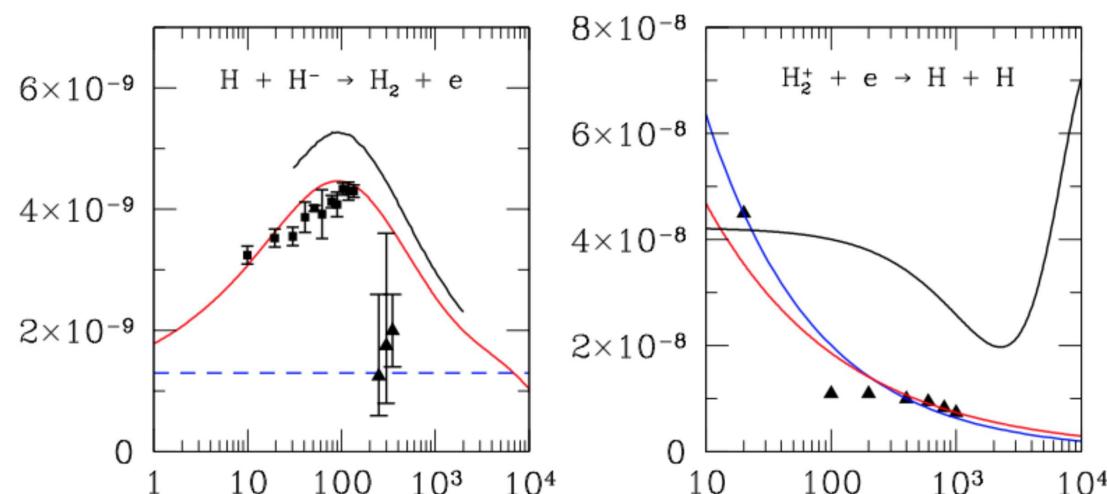
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■ Radiative association:



■ H⁻ formation:





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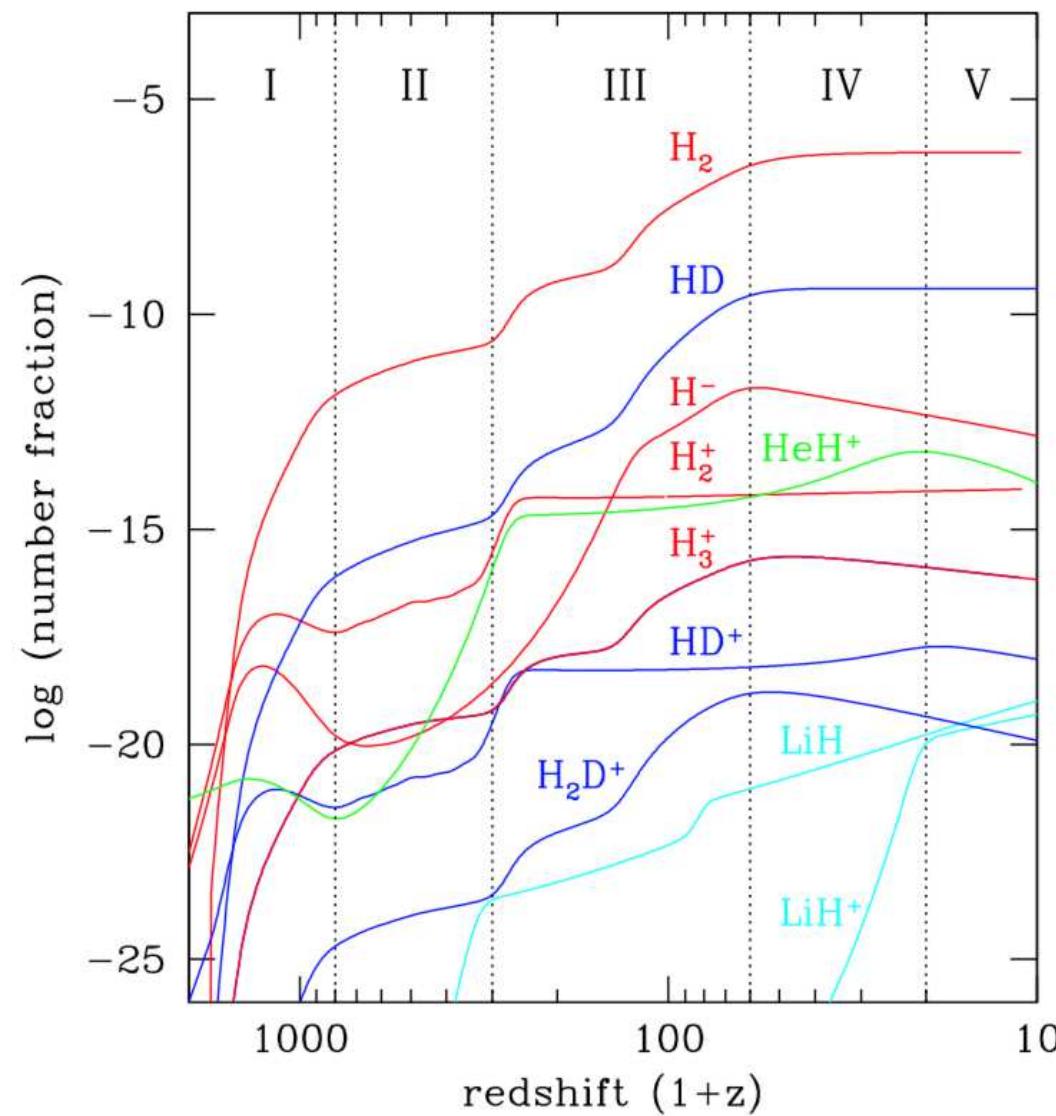
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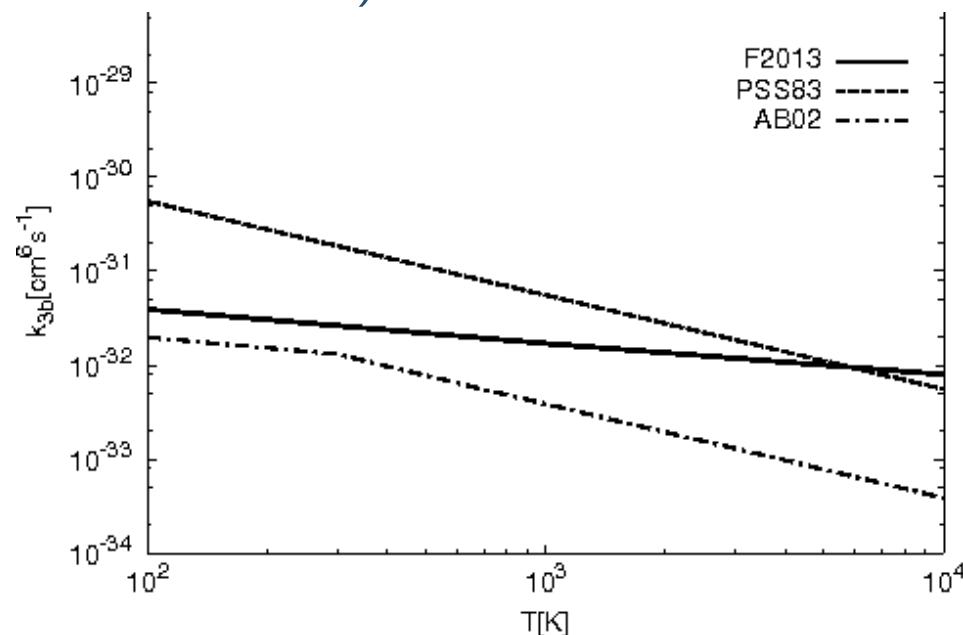
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- 3 body reactions (in collapsing clouds above 10^9 cm^{-3}):



- Rates based on reverse reaction rate measurement (collisional dissociation)





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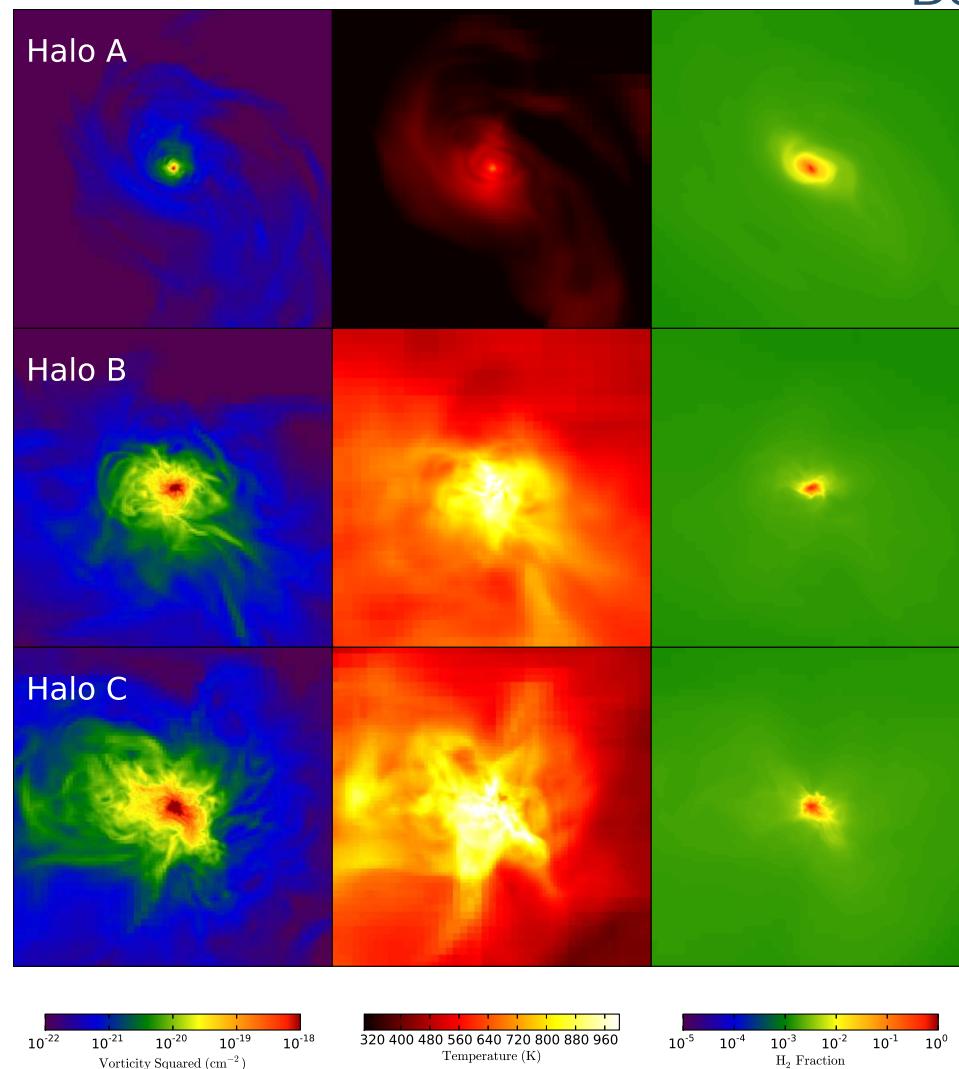
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Bovino et al., 2014, A&A,
561, 13



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Conclusion

- There is no conclusion!



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Conclusion

- There is no conclusion!
- Everywhere a huge amount of work is needed:
 - ◆ Theoretical
 - ◆ Experimental
 - ◆ Models
 - ◆ Observations



Conclusion

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Conclusion

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 - ◆ Theoretical
 - ◆ Experimental
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 - ◆ Observations
- Three papers:
 - ◆ Bron, E. et al., 2014, A&A, 569, 100
 - ◆ Sternberg, A., Le Petit, F. et al., 2014, ApJ, 790, 10
 - ◆ Galli, D., Palla, F., 2013, ARA&A, 51, 163