Quantum dynamics study of rate constant for a reactive collision of astrophysical interest :

the $D^+ + H_2$ reaction

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Collaborators/Fundings

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 $\underline{\text{Collaborators}}$:

• P. Honvault (ICB, Dijon, France)

• T. Gonzalez-Lezana (CSIC, Madrid, Spain) Fundings :



in a collaborative proposal with F. Lique* and A. Faure (2012-2014)

Deuterium Chemistry for ISM $D^+ + H_2 \rightarrow HD + H^+$

represents the major source of HD in diffuse interstellar clouds Dalgarno, Weisheit & Black, Ap.Lett., 14,77 (1973)

A&A 430, 967-977 (2005) DOI: 10.1051/0004-6361:20041589 © ESO 2005 Astronomy Astrophysics

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Deuterated molecular hydrogen in the Galactic ISM

New observations along seven translucent sightlines

S. Lacour^{1,2,3}, M. K. André¹, P. Sonnentrucker², F. Le Petit^{4,5}, D. E. Welty⁶, J.-M. Desert¹, R. Ferlet¹, E. Roueff⁵, and D. G. York⁶

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Addreck Warren of one design semantimum of the ID molecule in the intercenting as severely IT Galacia care. The final relation of the form of the final relation of the form of the final relation of

 \Rightarrow a low temperature rate constant k(T) is thus needed in order to get a correct estimation of H₂/HD ratio in the interstellar medium

and for Primordial Universe



- just after the recombination era, atomic and molecular process occur and play a role in the formation of first stars
- first structures are efficiently formed thanks to the molecular cooling of proto-molecular clouds
- H₂ and HD are the most efficient cooling molecules thus an accurate know ledg of their molecular abundance is required

and for Primordial Universe

THE ASTROPHYSICAL JOURNAL, 685:8–20, 2008 September 20

THE IMPACT OF HD COOLING ON THE FORMATION OF THE FIRST STARS

IAN D. MCGREER AND GREG L. BRYAN Department of Astronomy, Columbia University, Pupin Physics Laboratories, New York, NY 10027 Received 2008 February 25, accepted 2008 May 23

ABSTRACT

We use numerical simulations to investigate the importance of HD formation and cooling on the first generation of metal-free stars in a ACDM cosmology. We have implemented and tested non-equilibrium HD chemistry in an adaptive mesh refinement simulation code and applied it to two situations. (1) It is first applied to the formation of $0^{-1}0^{16}$, $M_{\rm ch}$ halos). We show, in agreement with previous work, that HD cooling is of only marginal importance for most halos, however, we find that for however any halos, with misses a few times 10⁴ M_c-HD cooling and equal or surpass the H_c cooling rate. This leads to appulation of stars formed in halos with effective HD cooling is to explore the impact of halos dominated by H_c cooling (2) In the second part of the paper, we ionize the halos in order to septore the impact of HD cooling is the presence of an ample population of presence of a surple population of stars of the CMB. We find HD cooling in the greater of halos dominated by H_c cooling in the accretion rate to the level of the CMB. We find that HD cooling dominates again to the presence of an ample population of stars of the more than the HD cooling dominates again to the elevent the finance that HD cooling dominates again to the presence of the ample to the level of the index to the HD cooling in the density range 10⁶-10⁶ cm⁻¹, but above this density the elevent to the provide the index to the HD cooling in the density range 10⁶-10⁶ cm⁻¹, but above this density, therefore we aggue that HD cooling in houring the secretion rate no the provide the index to the HD cooling is the H₂ cooling in the population of significanty) to were mass stars.

Subject headings: cosmology: theory — galaxies: formation — galaxies: high-redshift methods: n-body simulations — stars: formation

Online material: color figures

- HD is less abundant than H_2 ([D]/[H] $\simeq 2.10^{-5}$)
- but has a dipole moment ($\mu \simeq 8.3 \times 10^{-4}$ Debyes) and thus its cooling can equal or surpass the H₂ cooling rate

Chemical Network

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Galli and Palla. Astron. Astrophys., **335**, 403, 1998 Kinetic equation :

 $\frac{dn_x}{dt} = k_x^{form}(T_m)n_jn_w - k^{dest}(T_m)_xn_xn_u - \xi_x^{dest}(T_r)n_x + \xi_x^{form}(T_r)n_z + \dots$

Non-LTE effect

(日)

In many studies, molecular level populations are assumed to follow a Maxwell-Boltzmann law (LTE) but ...

• strong non-LTE effect have been shown for $H_2(v)$ in the early universe at low z



Coppola et al. ApJ. Series, **193**, 7, 2011

• several order of magnitude for the abundance of H_2^+ in the photodissociation $H_2^+ + h\nu \rightarrow H + H^+$ Galli and Palla. Astron. Astrophys., **335**, 403, 1998

State-to-State resolved kinetic

The population of a molecule should be solved self-consistently :

$$\frac{dn_{x,i}}{dt} = k_{x,i}^{form}(T_m)n_{y,k}n_{w,l} - k^{dest}(T_m)_x n_{x,i}n_{u,m}
- \xi_x^{dest}(T_r)n_{x,i} + \xi_x^{form}(T_r)n_{z,o} +
+ \sum_{j \neq i} n_{x,j}[A_{x,ji} + B_{x,ji}J_{ji}(T_r) + C_{x,ji}(T_m)]
- n_{x,i} \sum_{j \neq i} [A_{x,ij} + B_{x,ij}J_{ij}(T_r) + C_{x,ij}(T_m)]$$

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State-to-State resolved kinetic

FUTURE ISSUES

Major progress has been achieved by the inclusion of state-resolved chemical reaction rates for H₂ and H₂⁺, which are formed out of equilibrium in highly excited vibrational levels and then decay radiatively to the fundamental state. It is desirable to extend this approach to other species like HD, HD⁺, and HeH⁺. Further improvement comes from recent fully-quantal ab initio calculations of reaction cross sections for He and Li chemistry specifically performed for applications to the early Universe. Extending these calculations to other species/processes is an obvious need. It is necessary to complement these theoretical studies with laboratory experiments, especially at sub-electron volt energies.

Galli and Palla, Annu. Rev. Astron. Astrophys., **51**, 163 (2013) \Rightarrow state-resolved rate constant are required for early universe! \Rightarrow ab initio results at the state-of-the-art are very useful!

Experimental data for the HD formation

• First experimental study :

THE JOURNAL OF CHEMICAL PHYSICS VOLUME 34, NUMBER 9 1 MAY 1971

Investigation of Back-Scattering in the D^+/H_2 Reaction System^{*}

M. G. HOLLIDAY, † J. T. MUCKERMAN, AND L. FRIEDMAN Chemistry Department, Brookhaven National Laboratory, Upton, New York 11973 (Received 22 December 1970)

A collision chamber designed for efficient collection of product ions was used to determine the extent of back scattering in the reactions of D^+ and H_2 to form H^+ and HD^+ . No significant amount of back scattering in the laboratory system was detected.

• singles cross section values :

We also determined absolute cross sections (in the absence of a repeller field) at a laboratory energy of 7.8 eV (3.9 eV c.m.) for the reactions $D^++H_2 \rightarrow H^+ + HD$ and $D^++H_2 \rightarrow HD^+ + H$ using the procedure described in our previous paper. The values obtained were 0.26 Å² and 0.57 Å² for H⁺ and HD⁺ formation, re-

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... but experimental data obtained at room temperature thus not really useful for astrochemical models

Experimental data for the HD formation

• Second experimental study of k(T) at 80 K :

THE ASTROPHYSICAL JOURNAL, 183:L25-L26, 1973 July 1 © 1973. The American Astronomical Society. All rights reserved. Printed in U.S.A.

MEASUREMENT OF THE RATE CONSTANT FOR THE REACTION $\mathrm{D^+} + \mathrm{H_2} \! \rightarrow \! \mathrm{HD} + \mathrm{H^+}$

F. C. FEHSENFELD, D. B. DUNKIN, E. E. FERGUSON, AND D. L. ALBRITTON Aeronomy Laboratory, NOAA Environmental Research Laboratories, Boulder, Colorado *Received 1073 May 3*

ABSTRACT

It has recently been pointed out that the reaction $D^+ + H_2 \rightarrow HD + H^+$ can lead to enhanced HD/H₂ ratios in the interstellar medium if the rate constant is sufficiently large. We have measured the rate constant and found it to be 1.0×10^{-9} cm³ s⁻¹, and independent of temperature from 80° to 278° K.

Subject headings: interstellar matter - molecules

I. INTRODUCTION

Recent observations of the abundance of HD in interstellar molecules (Spitzer et al. 1973) have led to a keen interest in possible isotopic fractionation processes which may occur. It has been brought to our attention by Dalgarno (private communication; also Dalgarno, Black, and Weisheit et al. 1973) and by Watson (private communicanication; also Watson 1973) that a potentially important process is the exothermic ion-molecule reaction

$$D^+ + H_2 \rightarrow HD + H^+ + 0.039 \text{ eV},$$
 (1)

... but no temperature dependency of k(T) available from this work

Experimental data for the HD formation

• More recent experience :



Planetary and Space Science

N Plantey ad Spec Science (9909) (997-1997 (997) Deuterium fractionation in gas-phase reactions measured in the laboratory

D. Gerlich*, S. Schlemmer

Department of Physics, Technische Universität Chemnitz, 09107 Chemnitz, Germany Received 21 January 2002; received in revised form 15 April 2002; accepted 17 Jane 2002



Fig. 1. Energy dependence of the rate coefficients for the exothermic proton-dentor excetance $D^{-1} + H_{-} + H^{-1} + H^{-1}$ measured with different instruments (guided ion beam: GIB 74: Ocls and Teloy (1974), VT-SITF SI: Henchman et al. (1981), DBITF SZ: Villinger et al. (1982), GIB 92 and MB 92: (Corifich (1992)). The heavy line shows a rate coefficient, calculated with the MDB (most dynamically based) statistical theory by Gerlich (1982). Errors of the various experiments are indicated schematically as han or crosses.

First Principles calculations

In both dynamical approaches (TIQM/SQM) we have used the recent accurate H_3^+ PES of Velilla *et al.* L. Velilla, B. Lepetit, A. Aguado, J.A. Beswick and M. Paniagua, J.

Chem. Phys. 129, 084307 (2008) Collinear Representation of the H_3^+ potential :



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Theoretical method : TIQM

Hyperspherical formulation of the close-coupling approach

$$\sigma_{v'j',vj}(E) = \frac{\pi}{k_{vj}^2} \frac{1}{2j+1} \sum_{J\Omega'\Omega} (2J+1) \left| S_{v'j'\Omega',vj\Omega}^J(E) \right|^2$$

and the Differential Cross Section (DCS) :

$$\sigma_{v'j',vj}(\theta, E) = \frac{1}{8\mu E_c(2j+1)} \times \sum_{\Omega'\Omega} \left| \sum_J (2J+1) d^J_{\Omega\Omega'}(\pi-\theta) \right| \\ \times T^J_{v'j'\Omega',vj\Omega}(E) \right|^2$$

P. Honvault and J.-M. Launay, in Theory of Chemical Reaction Dynamics, edited by A. Lagana and G. Lendvay (Kluwer, Dordrecht, The Netherlands, 2004), Vol. 145, pp. 187215.

Adiabatic channels

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- For J = 0, more than 200 channels
- For $J \neq 0$, more than 1930 channels

Theoretical method : SQM

Assumption of a complex-forming dynamics of the process in which the state-to-state probability can be expressed as :

$$\mid S^{J}_{v'j'\Omega',vj\Omega}(E)\mid^{2}\simeq \frac{p^{J}_{v'j'\Omega'}(E)p^{J}_{vj\Omega}(E)}{\sum_{v''j''\Omega''}p^{J}_{v''j''\Omega''}(E)}$$

which can be used to determine the ICS $\sigma_{v'j',vj}(E)$. The DCS are approximated by :

$$\sigma_{v'j',vj}^{SQM}(\theta,E) = \frac{1}{8\mu E_c(2j+1)} \times \sum_{\Omega'\Omega} \sum_J (2J+1)^2 \left[d_{\Omega\Omega'}^J (\pi-\theta)^2 + d_{\Omega\Omega'}^J (\theta)^2 \right] \left| S_{v'j'\Omega',vj\Omega}^J (E) \right|^2$$





FIG. 1. Reaction probabilities for the $D^++H_2(v=0, j=0) \rightarrow H^++HD$ reaction at J=0. TIQM results (black line) are compared with statistical predictions (red line).

T. Gonzalez-Lezana, Y. Scribano and P. Honvault, J.Chem.Phys. **139**, 054301 (2013) $\langle \Box \rangle + \langle \Box \rangle +$

Integrated cross sections $\sigma(E)$



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Integrated cross sections $\sigma(E)$



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ICS : TIQM vs SQM

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FIG. 2. Integral cross sections for the $D^++H_2(v = 0, j = 0) \rightarrow H^++HD$ reaction. TIQM results (black line) are compared with statistical predictions in which only even (red lines)/all (blue lines) rotational states of $H_2(v = 0, j)$ are taken into account. The inset displays the same results on a log scale for the sake of clarity at the low energy regime. See text for details.

ICS : TIQM vs SQM

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FIG. 3. Same as Fig. 2 for D⁺+H₂(v = 0, j = 1) \rightarrow H⁺+HD reaction. SQM results are in this case for odd (red lines)/all (blue lines) rotational H₂(v = 0, j) states.

ICS : TIQM vs SQM



FIG. 4. State-to-state rate coefficients in terms of E_c for the D⁺+H₂(v = 0, j = 0) \rightarrow H⁺ +HD(v' = 0, j' = 0 - 2) reaction.

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Energy-Rate constant k(E)



P. Honvault and Y. Scribano, J.Phys.Chem. A 117, 9778 (2013)

- Very good agreement with experimental data of D. Gerlich
- Only a slight decrease when we go to low collisional energy

Rotational distribution





SQM calculations provides a reasonably good counterpart for the TIQM rotational distribution

Temperature-Rate constant k(T)

$$k(T) = \frac{1}{k_B T} \left(\frac{8}{\pi \mu k_B T}\right)^{1/2} \int_0^\infty \sigma(E_c) e^{-E_c/k_B T} E_c dE_c$$



T. Gonzalez-Lezana, Y. Scribano and P. Honvault, J. Phys. Chem. A, 2014, 118 (33), pp 6416-6424 An Analytical formula was fitted for KIDA

How to get the convergence?



FIG. 9. Convergence test of the cross sections for $D^+ +H(v = 0, j = 0)$ obtained with the SQM approach for different values of R_{max} . TIQM results⁴⁹ are shown for comparison.

T. Gonzalez-Lezana, P. Honvault and Y. Scribano, J.Chem.Phys. **139**, 054301 (2013)

- For low temperature (less than 10 K) the propagation should be done until a large asymptotic matching distance
- or use a more efficient propagation scheme (see Launay and Lara development of HYP3D)

Perspectives

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- Perform the complementary study of HD+H⁺ \rightarrow D⁺+H₂ (TIQM/SQM)
- Investigate the role of vibrationally excited $H_2(v)$ and HD(v') in the rate constant (SQM/TIQM)
- Calculations for higher temperature than 100 K for early universe application but ...

 $D^+ + H_2 \rightarrow HD + H^+$ $\rightarrow HD^+ + H$ (charge transfer)

- use of more efficient methods for such higher collisional energy range (such as TDWP method). see HeH⁺+H reaction, DeFazio, 2014
- use the state-to-state rate constant in a chemical model to evaluate the HD abundance for low temperature

Aknowledgements

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Thank you for your attention!